

Figure 1.10: This graph shows the Atomic Radius (distance from the center of the nucleus to the valence (outer) level of electrons) vs Atomic Number, displaying the trend in atomic radius for the first 40 elements in the periodic table.

**Description:** Each successive period is displayed in a different section of the atomic number (x-axis), ex. 1-10, 11-20, 21-30, etc. As the atomic number increases within a period, the atomic radius decreases. To put simply, the peaks represent group 1 or alkali metals and the troughs represent group 18 or halogens.

**DISCUSSION QUESTIONS:** (Please answer on a separate document/paper)

#### PART ONE:

- All the elements found at the peaks of your graph should fall into the same group.
- a. What is the group number?

Group I

b. What is the name of this group? Alkali metals

2. All the elements found at the troughs of your graph should fall into the same group.

A. What is the group number?

Group 18

# **Periodic Trends A Graphical Analysis Graph Answers**

**J Dewey** 

**Periodic Trends A Graphical Analysis Graph Answers:** 

## Periodic Trends A Graphical Analysis Graph Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Periodic Trends A Graphical Analysis Graph Answers**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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