

Step 8. Periodic table trends

Electronegativity

Ionization energy

He

Atomic radius

The periodic table displays various elements with their symbols and atomic numbers. Trends are indicated by colored arrows:

- Electronegativity:** Indicated by a black arrow pointing right across the top.
- Ionization energy:** Indicated by a green arrow pointing right across the top and a green arrow pointing up on the right side.
- Atomic radius:** Indicated by a red arrow pointing down on the left side and a red arrow pointing left across the bottom.
- Non-metallic properties:** Indicated by a purple arrow pointing up and to the right.
- Metallic properties:** Indicated by a blue arrow pointing down and to the right.

Ionization Energy

Electronegativity

Atomic radius

An arrow head shows the INCREASE of the property

Example: comparing K atom with Br atom

- Atomic radius of K is HIGHER than atomic radius of Br
- Ionization energy of K is LOWER than ionisation energy of Br
- Metallic properties of K are HIGHER than metallic properties of Br



Periodic Trends Key

Mike Jess

A red circular graphic with a gradient, appearing as a stylized arrow or a light effect, positioned to the right of the name bar.

Periodic Trends Key:

Periodic Trends Key Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Periodic Trends Key**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

<https://crm.avenza.com/files/Resources/HomePages/Pour%20Quelques%20Doigts.pdf>

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