UmSprint

Electronegativity

Ionization energy



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Atomic radius

An arrow head shows the INCREASE of the property

Example: comparing K atom with Br atom

- . Atomic radius of K is HIGHER than atomic radius of Br
- · Ionization energy of K is LOWER than ionisation energy of Br
- Metallic properties of K are HIGHER than metallic properties of Br

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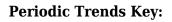
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Periodic Trends Key

Mike Jess



Periodic Trends Key Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has be more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Periodic Trends Key**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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