

# Part A

Answer all questions in this part.

Directions (1–30): For each statement or question, record on your separate answer sheet the number of the word or expression that, of those given, best completes the statement or answers the question. Some questions may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry.

1 Which statement describes the charge of an electron and the charge of a proton?

- (1) An electron and a proton both have a charge of +1.
- (2) An electron and a proton both have a charge of -1.
- (3) An electron has a charge of +1, and a proton has a charge of -1.
- (4) An electron has a charge of -1, and a proton has a charge of +1.

2 Which subatomic particles are found in the nucleus of an atom of beryllium?

- (1) electrons and protons
- (2) electrons and positrons
- (3) neutrons and protons
- (4) neutrons and electrons

3 The elements in Period 4 on the Periodic Table are arranged in order of increasing

- (1) atomic radius
- (2) atomic number
- (3) number of valence electrons
- (4) number of occupied shells of electrons

4 Which phrase describes two forms of solid carbon, diamond and graphite, at STP?

- (1) the same crystal structure and the same properties
- (2) the same crystal structure and different properties
- (3) different crystal structures and the same properties
- (4) different crystal structures and different properties

5 Which element has six valence electrons in each of its atoms in the ground state?

- (1) Se
- (2) As
- (3) Kr
- (4) Ga

6 What is the chemical name for  $\text{H}_2\text{SO}_3(\text{aq})$ ?

- (1) sulfuric acid
- (2) sulfurous acid
- (3) hydrosulfuric acid
- (4) hydrosulfurous acid

7 Which substance is most soluble in water?

- (1)  $(\text{NH}_4)_3\text{PO}_4$
- (2)  $\text{Cu}(\text{OH})_2$
- (3)  $\text{Ag}_2\text{SO}_4$
- (4)  $\text{CaCO}_3$

8 Which type of bonding is present in a sample of an element that is malleable?

- (1) ionic
- (2) metallic
- (3) nonpolar covalent
- (4) polar covalent

9 Which atom has the greatest attraction for the electrons in a chemical bond?

- (1) hydrogen
- (2) oxygen
- (3) silicon
- (4) sulfur

10 Which type of reaction involves the transfer of electrons?

- (1) alpha decay
- (2) double replacement
- (3) neutralization
- (4) oxidation-reduction

11 A 10.0-gram sample of nitrogen is at STP. Which property will increase when the sample is cooled to 72 K at standard pressure?

- (1) mass
- (2) volume
- (3) density
- (4) temperature

12 Which element is a gas at STP?

- (1) sulfur
- (2) xenon
- (3) potassium
- (4) phosphorus

# Nys Earth Science Regents June 2012 Answers

**K Payea**



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Let's Review: Earth Science Edward J. Denecke, 2009-09-01. Designed especially to help prepare students taking the New York State Regents Examination, this book makes a valuable supplementary text for high school level Earth Science classes throughout the country. Organized into three main study units: astronomy, meteorology, and geology, this book provides extensive subject review material with updated questions and answers. It also includes one recently given full-length Regents Exam in Earth Science.

Barron's Regents Exams and Answers David Berey, 1985      **Earth Science** Perry Cohen, 1965      **Barron's Regents Exams and Answers** Barron's Educational Series, Inc., David Berey, Stanley Henry Kaplan, Maurice Bleifeld, 1969      **Earth Science: the Physical Setting** Paola Santagostino, Prentice Hall (School Division), 2005. Focusing on the Earth Science content tested on the Regents Examination, this thorough review guide contains extensive vocabulary, review questions, and Memory Jogger and Digging Deeper features. Hundreds of practice questions organized in the Regents Examination format help students familiarize themselves with look and feel of the actual exam.

Regents Exams and Answers: Earth Science--Physical Setting Revised Edition, 2019      **Regents Exams and Answers: Earth Science** Edward J. Denecke, 2017-11-01. Always study with the most up-to-date prep. Look for Regents Exams and Answers Earth Science Physical Setting 2020 ISBN 978 1 5062 5399 2 on sale January 07 2020. Publisher's Note: Products purchased from third-party sellers are not guaranteed by the publisher for quality, authenticity, or access to any online entitlements included with the product.

**Traditional Vs. Experiential** Subhas Mohan, 2015. This study explores the differences in student achievement on state standardized tests between experiential learning and direct learning instructional methodologies. Specifically, the study compares student performances in Expeditionary Learning schools, which is a Comprehensive School Reform model that utilizes experiential learning, to their counterparts or peer schools that utilize traditional instructional methodology. This study employs a quasi-experimental quantitative design. Student test scores and individual answer responses on the New York State Regents Living Environment Earth Science and Chemistry science exams from the June 2012 and June 2013 administration were analyzed for both the Experiential Learning and the Traditional Learning groups. Descriptive as well as inferential statistical analyses were performed on the data to determine the differences in students' mean scores on various attributes of the test as well as differences in achievement levels for various subgroups of the sample. The study found that the Experiential Learning group

had statistically significant greater mean scores in overall performance on Living Environment Earth Science and Chemistry exams than the Traditional Learning group. The study also found that students in the Experiential Learning group outperformed their peers on critical thinking questions on all three science exams combined as well as on the Earth Science laboratory practicum exam. Additionally, the study found that English Language Learners (ELLs) and students with an Individual Education Plan (IEP) in the Experiential Learning group significantly outperformed their counterparts in the Traditional Learning group. This study concludes that Experiential Learning should be leveraged as a comprehensive school reform model to increase students' overall performance on state standardized science examinations. This study also concludes that Experiential Learning should be capitalized upon to increase student development of critical thinking skills as well as for students with special academic needs. Finally, this study provides further evidence that district and school leaders should strongly consider Experiential Learning as a comprehensive school reform model, a viable option to enable student achievement.

**Earth Science** Jack Rudman, 1980

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