

Gas laws practice test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ____ 1. Pressure is the force per unit
a. volume. c. length.
b. surface area. d. depth.
- ____ 2. Why does a can collapse when a vacuum pump removes air from the can?
a. The inside and outside forces balance out and crush the can.
b. The unbalanced outside force from atmospheric pressure crushes the can.
c. The atmosphere exerts pressure on the inside of the can and crushes it.
d. The vacuum pump creates a force that crushes the can.
- ____ 3. If the height of mercury in a barometer at 0°C is less than 760 mm Hg, then
a. the atmospheric pressure is less than standard atmospheric pressure.
b. the atmospheric pressure is greater than standard atmospheric pressure.
c. the atmospheric pressure is equal to standard atmospheric pressure.
d. the atmospheric pressure cannot be determined.
- ____ 4. Convert the pressure 0.840 atm to mm Hg.
a. 365 mm Hg c. 638 mm Hg
b. 437 mm Hg d. 780 mm Hg
- ____ 5. Convert the pressure 1.30 atm to kPa.
a. 2 kPa c. 132 kPa
b. 115 kPa d. 245 kPa
- ____ 6. Standard temperature is exactly
a. 100°C. c. 0°C.
b. 273°C. d. 0 K.
- ____ 7. Three samples of gas each exert 740. mm Hg in separate 2 L containers. What pressure do they exert if they are all placed in a single 2 L container?
a. 247 mm Hg c. 1.48×10^3 mm Hg
b. 740 mm Hg d. 2.22×10^3 mm Hg
- ____ 8. If the temperature remains constant, V and P represent the original volume and pressure, and V' and P' represent the new volume and pressure, what is the mathematical expression for Boyle's law?
a. $P'V = V'P$ c. $VP = V'P'$
b. $VV' = PP'$ d. $V' = \frac{VP}{P'}$
- ____ 9. A sample of oxygen occupies 560. mL when the pressure is 800.00 mm Hg. At constant temperature, what volume does the gas occupy when the pressure decreases to 700.0 mm Hg?
a. 80.0 mL c. 600. mL
b. 490. mL d. 640. mL
- ____ 10. If the temperature of a fixed quantity of gas decreases and the pressure remains unchanged,
a. its volume increases. c. its volume decreases.
b. its volume is unchanged. d. its density decreases.

Sample Test Questions On Gas Laws

Arshad Iqbal



Sample Test Questions On Gas Laws:

Unveiling the Magic of Words: A Overview of "**Sample Test Questions On Gas Laws**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Sample Test Questions On Gas Laws**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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Table of Contents Sample Test Questions On Gas Laws

1. Understanding the eBook Sample Test Questions On Gas Laws
 - The Rise of Digital Reading Sample Test Questions On Gas Laws
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Test Questions On Gas Laws
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Test Questions On Gas Laws
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Test Questions On Gas Laws
 - Personalized Recommendations
 - Sample Test Questions On Gas Laws User Reviews and Ratings
 - Sample Test Questions On Gas Laws and Bestseller Lists

5. Accessing Sample Test Questions On Gas Laws Free and Paid eBooks
 - Sample Test Questions On Gas Laws Public Domain eBooks
 - Sample Test Questions On Gas Laws eBook Subscription Services
 - Sample Test Questions On Gas Laws Budget-Friendly Options
6. Navigating Sample Test Questions On Gas Laws eBook Formats
 - ePub, PDF, MOBI, and More
 - Sample Test Questions On Gas Laws Compatibility with Devices
 - Sample Test Questions On Gas Laws Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Test Questions On Gas Laws
 - Highlighting and Note-Taking Sample Test Questions On Gas Laws
 - Interactive Elements Sample Test Questions On Gas Laws
8. Staying Engaged with Sample Test Questions On Gas Laws
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sample Test Questions On Gas Laws
9. Balancing eBooks and Physical Books Sample Test Questions On Gas Laws
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Test Questions On Gas Laws
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Sample Test Questions On Gas Laws
 - Setting Reading Goals Sample Test Questions On Gas Laws
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sample Test Questions On Gas Laws
 - Fact-Checking eBook Content of Sample Test Questions On Gas Laws
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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