

ICSE Class 10 Chemistry Question Paper Solution 2016

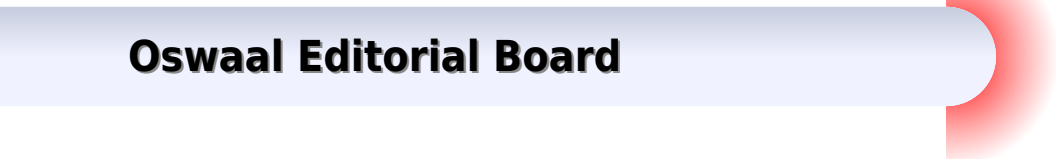
CHEMISTRY SCIENCE Paper – 2

Question 1

- (a) Fill in the blanks with the choices given in brackets. [5]
- (i) Metals are good _____ (*oxidizing agents / reducing agents*) because they are electron _____ (*acceptors/ donors*).
 - (ii) Electrovalent compounds have _____ (*high / low*) melting points.
 - (iii) Higher the pH value of a solution, the more _____ (*acidic / alkaline*) it is.
 - (iv) _____ (*AgCl / PbCl₂*), a white precipitate is soluble in excess NH_4OH .
 - (v) Conversion of ethene to ethane is an example of _____ (*hydration / hydrogenation*).
- (b) Choose the *correct answer* from the options given below: [5]
- (i) An element with the atomic number 19 will most likely combine chemically with the element whose atomic number is:
 - A. 17
 - B. 11
 - C. 18
 - D. 20
 - (ii) The ratio between the number of molecules in 2g of hydrogen and 32g of oxygen is:
 - A. 1:2
 - B. 1:0.01
 - C. 1:1
 - D. 0.01:1 [Given that H=1, O=16]
 - (iii) The two main metals in *Bronze* are:
 - A. Copper and zinc
 - B. Copper and lead
 - C. Copper and nickel
 - D. Copper and tin
 - (iv) The particles present in *strong electrolytes* are:
 - A. only molecules
 - B. mainly ions

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