Density Worksheet Physical Science D=m/V

Substance	Density (g/cm²)	Substance	Denwity (g/ems)
Oxygen.	0.00033	Aluminum	9.70
Hydrogen	0.000084	Iron	7.87
Ethanol	0.786	Copper	8.96
Benzene	0.880	Silver	10.5
Water	1.000	Lead	31-34
Magnesium	3-74	Mercury	33.6
Salt (sedium chloride)	2.16	Gold	19.30

 The ratio of an object's mass to its ______ is called the devisity of the object.

 A kilogram of lead occupies a much smaller volume than a kilogram of water, because has a much higher density.

 For the masses and volumes indicated, calculate the density in grams per cubic centimeters.

a. mass = 453 g; volume = 225 cm3

b. mass = 5.0 g; volume = 10.0 cm3

c. mass = 26.1 g; volume = 2.0 mL

4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.

A cube of metal weighs 1450 g and displaces 542 mL of water when immersed.Calculate the density of the metal.

Physical Science If8767 Density

J Spring

Physical Science If8767 Density:

This book delves into Physical Science If8767 Density. Physical Science If8767 Density is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and indepth insights into Physical Science If8767 Density, encompassing both the fundamentals and more intricate discussions.

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