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# Class 10 Physical Science

Chemistry Formal Experiment: Using flame colours to identify unknown metal ions

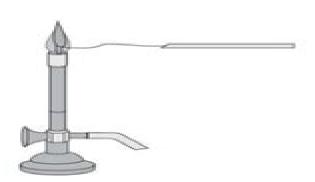
	August 2024
	60 Marks
Full Name:	1 hour

#### Introduction:

The flame test is used to visually determine the identity of an unknown metal or metalloid, as an element or in a compound, based on the characteristic colour the salt turns the flame of a burner. When an electron in an excited state (higher energy) returns to its ground state (lower energy), an atomic spectrum is produced. A specific quantity of energy is associated with each electron, and when an electron returns to a lower energy level, a specific quantity of energy is released. The energy is released as electromagnetic radiation. Certain wavelengths will be visible to the naked eye; other wavelengths can only be detected with instruments. In the activity below, we will investigate the colours emitted by certain metal atoms. Different colours indicate different wavelengths of visible radiation.

#### Materials:

NaCl(aq), KCl(aq); $CaCl_2.2H_2O(aq)$ , $CuCl_2.2H_2O(aq)$
copper powder, magnessum powder,
zinc powder, iron powder
four unknown salt solutions
Bunsen burner
Nichrome wire



# **Physical Science Formal Experiment Nomemo 2014**

**L** Manion

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#### **Table of Contents Physical Science Formal Experiment Nomemo 2014**

- 1. Understanding the eBook Physical Science Formal Experiment Nomemo 2014
  - The Rise of Digital Reading Physical Science Formal Experiment Nomemo 2014
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Physical Science Formal Experiment Nomemo 2014
  - Exploring Different Genres
  - $\circ\,$  Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Formal Experiment Nomemo 2014
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physical Science Formal Experiment Nomemo 2014
  - Personalized Recommendations
  - Physical Science Formal Experiment Nomemo 2014 User Reviews and Ratings

- Physical Science Formal Experiment Nomemo 2014 and Bestseller Lists
- 5. Accessing Physical Science Formal Experiment Nomemo 2014 Free and Paid eBooks
  - Physical Science Formal Experiment Nomemo 2014 Public Domain eBooks
  - Physical Science Formal Experiment Nomemo 2014 eBook Subscription Services
  - Physical Science Formal Experiment Nomemo 2014 Budget-Friendly Options
- 6. Navigating Physical Science Formal Experiment Nomemo 2014 eBook Formats
  - o ePub, PDF, MOBI, and More
  - Physical Science Formal Experiment Nomemo 2014 Compatibility with Devices
  - Physical Science Formal Experiment Nomemo 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Formal Experiment Nomemo 2014
  - Highlighting and Note-Taking Physical Science Formal Experiment Nomemo 2014
  - o Interactive Elements Physical Science Formal Experiment Nomemo 2014
- 8. Staying Engaged with Physical Science Formal Experiment Nomemo 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Formal Experiment Nomemo 2014
- 9. Balancing eBooks and Physical Books Physical Science Formal Experiment Nomemo 2014
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Formal Experiment Nomemo 2014
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Formal Experiment Nomemo 2014
  - Setting Reading Goals Physical Science Formal Experiment Nomemo 2014
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Formal Experiment Nomemo 2014
  - Fact-Checking eBook Content of Physical Science Formal Experiment Nomemo 2014
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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