

A chemical equation is a representation of a chemical reaction in which the reactants are written using chemical formulas, followed by an arrow.

Reactants Products

A coefficient represents the number of atoms or molecules of a substance that are involved in a chemical reaction. A coefficient is placed directly in front of a chemical formula.

A subscript is placed after the element symbol to indicate the number of atoms of that element in a molecule.

Why is it important to balance chemical equations?

Chemical equations must be balanced because matter is conserved. The number of atoms of each element must be the same on both sides of the equation.

What does the law of conservation of mass state?

The law of conservation of mass states that matter is neither created nor destroyed in a chemical reaction.

Week 8, Day 2

The number of protons in an atom cannot change (it is fixed) but a different number of neutrons can change.

The most stable electron configuration for atoms is when they have a complete set of valence electrons (the highest energy level is full).

How does a positive ion form from a neutral atom? What is another name for a positive ion?

When an atom loses one or more electrons, it becomes positively charged and is called a positive ion or cation.

How does a negative ion form from a neutral atom? What is another name for a negative ion?

When an atom gains one or more electrons, it becomes negatively charged and is called a negative ion or anion.



Atoms in Group 1A have 1 electron in their outer shell and Group 2A atoms have 2 electrons in their outer shell. To become more stable (like the noble gases), they easily give up their valence electrons. Group 1A atoms have 1 electron in their outer shell and easily lose it to become the more stable noble gas configuration of the next noble gas.



Atoms are formed by taking the first part of the element name and adding the suffix -ide. For example, sodium chloride, NaCl, is called a sodium chloride.



When sodium and chlorine atoms react, the sodium atom gives up the valence electron, becoming a cation (positively charged ion), and the chlorine atom gains the electron, becoming an anion (negatively charged ion).

The positively charged sodium ion and the negatively charged chloride ion are attracted to each other. The bond formed between the oppositely charged ions, holding the two ions together.

Physical Science Exploring Space Notetaking Guide

Answer

Xiaolong Qi



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