

Support

Customer Service

SoftBank Customer Center
Toll-free numbers
188-123456789
188-123456789

SoftBank Global Call Center

188-123456789
188-123456789
188-123456789

More Information

Handset User's Guide
Online English Manual
SoftBank Global Call Center
188-123456789
188-123456789

Call these numbers toll-free from landlines

Service Area	Service Center	Phone Number
Japan (Tokyo, Osaka, Nagoya, Fukuoka, etc.)	Service Center	188-123456789
Japan (Tokyo, Osaka, Nagoya, Fukuoka, etc.)	Service Center	188-123456789
Japan (Tokyo, Osaka, Nagoya, Fukuoka, etc.)	Service Center	188-123456789
Japan (Tokyo, Osaka, Nagoya, Fukuoka, etc.)	Service Center	188-123456789
Japan (Tokyo, Osaka, Nagoya, Fukuoka, etc.)	Service Center	188-123456789
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Japan (Tokyo, Osaka, Nagoya, Fukuoka, etc.)	Service Center	188-123456789
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SoftBank 8255H First Steps Guide

Emergency Numbers

Customer Service

More Information

Use English Interface

Place International Calls

Customer Handset Address

And more...



Safety Precautions

- Do not use the handset near flammable, explosive, etc. items.
- Do not use the handset near fire, heat, or other sources of high temperature.
- Do not use the handset near water, or other liquids.
- Do not use the handset near strong magnetic fields, or other sources of high magnetic field.
- Do not use the handset near strong electric fields, or other sources of high electric field.
- Do not use the handset near strong radio waves, or other sources of high radio frequency.
- Do not use the handset near strong electromagnetic waves, or other sources of high electromagnetic field.
- Do not use the handset near strong static electricity, or other sources of high static electric field.
- Do not use the handset near strong static magnetic fields, or other sources of high static magnetic field.
- Do not use the handset near strong static electric and magnetic fields, or other sources of high static electric and magnetic field.

- Do not use the handset near strong static electric and magnetic fields, or other sources of high static electric and magnetic field.
- Do not use the handset near strong static electric and magnetic fields, or other sources of high static electric and magnetic field.
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- Do not use the handset near strong static electric and magnetic fields, or other sources of high static electric and magnetic field.

Before Using Handset

Accessories

- Battery (2 pieces)



Using Software (Optional)



Using Card Installation

- Make a SIM Card with 10-step background

Insert Battery (Optional)

- Do not use the handset with damaged or worn-out battery.
- Do not use the handset with damaged or worn-out battery.
- Do not use the handset with damaged or worn-out battery.



Minimizing Mode

- Minimizing Mode (Optional)
- Minimizing Mode (Optional)
- Minimizing Mode (Optional)

Minimizing/Maximizing Minimizing Mode

- Minimizing Mode

Battery Installation

- Remove cover



- Insert battery



- Close cover



- Press and hold as shown

Using Mode

- 1. Press
- 2. Press
- 3. Press

Using Mode (Optional)

- 1. Press
- 2. Press
- 3. Press
- 4. Press
- 5. Press
- 6. Press
- 7. Press
- 8. Press
- 9. Press
- 10. Press

Charging

- 1. Press
- 2. Press
- 3. Press
- 4. Press
- 5. Press
- 6. Press
- 7. Press
- 8. Press
- 9. Press
- 10. Press



Security Codes

- 1. Press
- 2. Press
- 3. Press
- 4. Press
- 5. Press
- 6. Press
- 7. Press
- 8. Press
- 9. Press
- 10. Press

- 1. Press
- 2. Press
- 3. Press
- 4. Press
- 5. Press
- 6. Press
- 7. Press
- 8. Press
- 9. Press
- 10. Press

Specifications

Model	SoftBank 8255H
Display	2.4 inch, 262K colors
Camera	2.0 megapixel
Memory	128 MB
Operating System	Android 2.2
Charging Time	10 hours
Standby Time	10 hours

Model	SoftBank 8255H
Display	2.4 inch, 262K colors
Camera	2.0 megapixel
Memory	128 MB
Operating System	Android 2.2
Charging Time	10 hours
Standby Time	10 hours

Sharp 825sh User Guide

RJ Shavelson



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2 3 isotopes when the number of neutrons varies - Feb 15 2023

web atoms of the same element that differ in their numbers of neutrons are called isotopes many isotopes occur naturally usually one or two isotopes of an element are the most stable and common different isotopes of an element generally have the same physical and chemical properties because they have the same numbers of protons and electrons

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web jan 5 2016 explanation isotopes are atoms of the same element same atomic number i e same number of protons in the nucleus but different atomic masses different mass numbers due to different numbers of neutrons in the nucleus for example we may get an isotope of chlorine with 17 protons and 18 neutrons hence atomic mass 35

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web may 16 2018 explanation an atom is characterized as a particular element by the number of positively charged massive nuclear particles i e Z $Z=1$ we get hydrogen $Z=2$ we get helium $Z=3$ lithium $Z=47$ we get silver the nucleus can also contain

neutrons massive nuclear particles of zero charge

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web atoms of the same element with different numbers of neutrons are called isotopes close isotope atoms of an element with the same number of protons and electrons but different numbers of neutrons

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web oct 31 2023 an isotope is one of two or more species of atoms of a chemical element with the same atomic number and position in the periodic table and nearly identical chemical behavior but with different atomic masses and physical properties every chemical element has one or more isotopes

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web jul 18 2018 answer link due to the number of neutrons isotopes differ with respect to the number of neutrons their nuclei contain where neutrons are massive particles of zero charge a nucleus may contain variable numbers of neutrons and this gives rise to the phenomenon of isotopes

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web isotopes are distinct nuclear species or nuclides as technical term of the same chemical element they have the same atomic number number of protons in their nuclei and position in the periodic table and hence belong to the same chemical element but differ in nucleon numbers mass numbers due to different numbers of neutrons in

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little

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web chemistry questions and answers isotopes differ from each other in what ways a they have different numbers of

electrons outside the nucleus b they have different atomic number o c they have different numbers of neutrons in the nucleus
d more than one response is correct e they have different numbers of protons in the nucleus