














Molecular Geometry Chart

| # of Electron Groups | Number of Lone Pairs | Electron Pair Arrangement | Molecular Geometry | Approximate Bond Angles |
|----------------------|----------------------|---------------------------|---|-------------------------|
| 2 | 0 | linear |  | 180° |
| 3 | 0 | trigonal planar |  | 120° |
| | 1 | bent |  | <120° |
| 4 | 0 | tetrahedral |  | 109.5° |
| | 1 | trigonal pyramid |  | <109.5° (~107°) |
| | 2 | bent |  | <109.5° (~105°) |
| 5 | 0 | trigonal bipyramidal |  | 90°, 120° |
| | 1 | see-saw |  | <90°, <120° |
| | 2 | T-structure |  | <90° |
| | 3 | linear |  | 180° |
| 6 | 0 | octahedral |  | 90°, 90° |
| | 1 | square pyramidal |  | 90°, <90° |
| | 2 | square planar |  | 90° |

Shapes Of Molecules Worksheet If8766

Paul G. Mezey

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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