Memorandum arade 11 Pisciences

Physical Sciences Gr11/P2

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QUESTION 1

- 1.1 But of all
- 1.2 BYY
- 1.3 OBY
- 民ロンノ 1.4
- NY

- (2)
- (2)
- (2)
- (2)

QUESTION 2

2.1 The mass of one mole of a substance measured in g-mol⁻¹ < < (2 or 0) (20)

- 2.2
- Marking criteria (a) Any formula m = nM or N = nNA -
 - (b) Substitution of 3.114×10²⁴ and 6.02×10²³ into N = nN_A
 - (c) Substitution of 1 g-mol* Into m = nM <
 - (d) Mole ratio 3:6:1 -

C	H	0
$n = \frac{N}{N_A}$ $n = \frac{3.114 \times 10^{24}}{6.02 \times 10^{23}}$ $n = 5.173 \text{ mol}$	$n = \frac{m}{M}$ $n = \frac{10,345}{1 \checkmark}$ $n = 10,345 \text{ mol}$	n = 1,724 mol
$\frac{5,173}{1,724} = 3$	$\frac{10,345}{1,724} = 6$	$\frac{1,724}{1,724} = 1$
	3:6:1 ✓	
X=3-/	Y=6~	0=1~

Accept values indicated in the empirical formula: C₂H₆O

(7)

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