

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ relationship _____ equated _____

2. $40g = 40r + 4 \cdot 20g$

2. Magnesium nitride reacts with a single oxygen when the products of synthesis are at magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ from the _____ reaction in the _____

3. Problem

When a chemical substance reacts with a chemical system, a chemical reaction is produced

$$2 \cdot A_2 + 3g \cdot B \rightarrow 2AB_2$$

moles reaction

$$\begin{array}{cccc} 2 & A_2 & 3 & B \\ 2 & A_2 & 3 & AB_2 \\ 2 & B_2 & 3 & AB_2 \end{array}$$

2. If you start with 4 moles of A_2 , how many moles of AB_2 will be produced?

2. $2 \cdot A_2 + 3g \cdot B \rightarrow 2AB_2$ and $4g \cdot B \rightarrow 4 \cdot 20g$

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