

## **OCR SALTERS B CHEMISTRY- PAPER 3 PRACTICAL SKILLS 2024 QUESTIONS WITH CORRECT ANSWERS ALREADY PASSED AND VERIFIED**

**Weighing a solid - CORRECT ANSWER** 1. Zero the balance

2. Place weighing bottle/boat onto balance and add in the approximate required mass of solid
3. Accurately weigh solid and weighing bottle and record
4. Empty solid into glassware where it will be used
5. Accurately reweigh empty weighing bottle
6. Subtract recorded mass for empty weighing bottle from mass recorded for solid and weighing bottle

**Measuring a volume of a liquid using a pipette - CORRECT ANSWER** Pipettes used for accurately dispensing a fixed volume of a liquid, usually 10 or 25cm<sup>3</sup>

1. Ensure pipette is completely clean- rinse with water and then a small volume of the solution that will be pipetted
2. Dip pipette into solution and, using a pipette filler, draw enough liquid into the pipette until it is exactly the right level (read using bottom of meniscus at eye level)
3. Run liquid out of pipette into glassware until it stops. Touch the pipette on the side and remove (will be drop left in)

# Salters B F33june 2013 Paper

**Mike Jess**



**Salter's B F33 June 2013 Paper:**

## **Salters B F33june 2013 Paper** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the ability of words has be evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Salters B F33june 2013 Paper**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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sifat sifat gas mulia wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik molekul yg terdiri dari 1 atom ini erat kaitannya dengan struktur elektron oktet dan duplet dari gas mulia

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halogen memiliki 6 bilangan oksidasi sehingga mudah sekali berreaksi dengan unsur lainnya gas mulia noble

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pembuatan gas mulia gas helium mempunyai titik didih yang sangat rendah yaitu 268.8 K sehingga pemisahan gas helium dari gas alam dilakukan dengan cara pengembunan liquefaction sampai gas alam akan mencair sekitar 156 K dan gas helium terpisah dari gas alam pembuatan neon argon krypton dan xenon dilakukan dengan proses destilasi bertingkat

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golongan gas mulia dan halogen beberapa golongan unsur memiliki nama khusus seperti golongan VIII gas mulia dan VIIA halogen golongan tersebut memiliki nama khusus seperti sifat periodik sifat fisik dan sifat kimia khas sifat ke 2 golongan tersebut dapat membantu kita unsur unsur yang lain dalam 1 golongan semakin ke atas dan dalam

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