EXAMINATION DATA SHEET FOR THE PHYSICAL SCIENCES (CHEMISTRY)

TABLE 1 PHYSICAL CONSTANTS

NAME	SYMBOL	VALUE
Magnitude of charge on electron	е	1,6 × 10 ⁻¹⁹ C
Mass of an electron	me	9.1 × 10 ⁻³¹ kg
Standard pressure	₽ ⁶	1,01 × 10 ⁵ Pa
Motar gas volume at STP	V.,	22,4 dm ³ - mol-1
Standard temperature	T*	273 K (0 °C)
Avogadro's constant	Na	6,02 × 10 ²³ mol ⁻¹
Faraday's constant	F	96 500 C- mol-1

TABLE 2 CHEMISTRY FORMULAE

$n = \frac{m}{M}$	$n = \frac{N}{N_A}$	$n = \frac{\nu}{\nu_{\infty}}$
$c = \frac{n}{V}$ OR $c = \frac{m}{MV}$		0 ⁴] [OH ⁻] = 1 ×10 ⁻¹⁴ 25 °C (298 K)
q = It		ode - E ^θ anode
	400	ent – E ⁸ reducing agent

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Xiaolong Qi

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Physical Science P2 2014 Prelims**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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