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Liber the terms from the following list to fit each of the claes given below

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Clues	Vocabulary Terms
A type of bond that holds cations and anions together	SUDANNE CORNELIA I I I
A type of bond in which two atoms share a pair of tralence electrons	
A neutral group of atoms that are joined together by one or more covalent bonds	
A term describing a covalent bond in which electrons are not shared equally	
An ion that contains a covalently bonded group of atoms	
An ion with a negative charge	
An ion with a positive charge	
A notation that shows what elements a compound contains and the ratio of the atoms or ions of these elements in the compound	
Solids whose particles are arranged in a lattice structure	
A mixture of two or more elements, at least one of which is a metal.	
A type of bond that exists between a metal cation and the shared electrons that surround it	

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