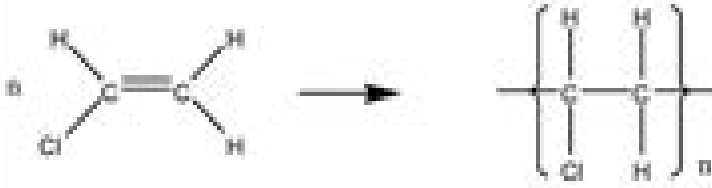


Question	Answer	Mark	Guidance
5 (a)	 Correct polymer with side links ✓ Balanced equation for formation of correct polymer - correct use of n in the equation and brackets ✓	2	Displayed formulae MUST be used to award each mark n on LHS can be at any height to the left of formula AND n on the RHS must be a subscript (essentially below the side link)
(b) (i)	$\text{CH}_3\text{CHCl} + 2\text{O}_2 \rightarrow \text{CO} + \text{CO}_2 + \text{HCl} + \text{H}_2\text{O} \checkmark$	1	ALLOW any other correctly balanced equation with the same reactants and products ALLOW $\text{C}_2\text{H}_3\text{Cl}$ for CH_3CHCl
(ii)	Sodium hydrogencarbonate neutralises HCl ✓	1	Assume that 'it' refers to sodium hydrogencarbonate but DO NOT ALLOW other chemicals e.g. sodium ALLOW NaHCO_3 is a base ALLOW forms a salt or sodium chloride or NaCl ALLOW equation to show formation of NaCl from NaHCO_3 and HCl even if not balanced. IGNORE reacts

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Table of Contents Ocr F322 June 2014 Unofficial Mark Scheme

1. Understanding the eBook Ocr F322 June 2014 Unofficial Mark Scheme
 - The Rise of Digital Reading Ocr F322 June 2014 Unofficial Mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr F322 June 2014 Unofficial Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr F322 June 2014 Unofficial Mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr F322 June 2014 Unofficial Mark Scheme
 - Personalized Recommendations
 - Ocr F322 June 2014 Unofficial Mark Scheme User Reviews and Ratings
 - Ocr F322 June 2014 Unofficial Mark Scheme and Bestseller Lists

5. Accessing Ocr F322 June 2014 Unofficial Mark Scheme Free and Paid eBooks
 - Ocr F322 June 2014 Unofficial Mark Scheme Public Domain eBooks
 - Ocr F322 June 2014 Unofficial Mark Scheme eBook Subscription Services
 - Ocr F322 June 2014 Unofficial Mark Scheme Budget-Friendly Options
6. Navigating Ocr F322 June 2014 Unofficial Mark Scheme eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr F322 June 2014 Unofficial Mark Scheme Compatibility with Devices
 - Ocr F322 June 2014 Unofficial Mark Scheme Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr F322 June 2014 Unofficial Mark Scheme
 - Highlighting and Note-Taking Ocr F322 June 2014 Unofficial Mark Scheme
 - Interactive Elements Ocr F322 June 2014 Unofficial Mark Scheme
8. Staying Engaged with Ocr F322 June 2014 Unofficial Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr F322 June 2014 Unofficial Mark Scheme
9. Balancing eBooks and Physical Books Ocr F322 June 2014 Unofficial Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr F322 June 2014 Unofficial Mark Scheme
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ocr F322 June 2014 Unofficial Mark Scheme
 - Setting Reading Goals Ocr F322 June 2014 Unofficial Mark Scheme
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr F322 June 2014 Unofficial Mark Scheme
 - Fact-Checking eBook Content of Ocr F322 June 2014 Unofficial Mark Scheme
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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