

Final Exam Study Guide

Module 7: Reading Strategies

1. What is active reading?
2. What is annotating?
3. What is a chapter summary?
4. What are context clues?
5. What is a glossary?
6. What is highlighting regarding reading strategies?
7. What is an index?
8. What are notes?
9. What is previewing?
10. What is skimming?
11. What is the table of contents?

Module 8: Critical Thinking Skills Fake News and Bias

12. If someone is biased, what does that mean?
13. What is fake news?
14. What is the purpose of fake news?
15. What are some ways you can tell that a piece of material is fake news?
16. When does social bias occur?
17. What is a confirmation bias?
18. How can assessing your own personal biases help you?
19. What are characteristics of a fake news article?
20. What are characteristics of a real news article?
21. What is clickbait?

Module 9: Test Taking Strategies

22. What kind of question should you use process of elimination on?
23. What is a process of elimination?
24. What is a Test Smart Plan?
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27. What are guided words?
28. If you have a passage to read with questions to follow, what should be done first?
29. What are unconditional words?

Module 10: Academic Integrity

30. What is the definition of a popular source? And give an example of a popular source.
31. What is the definition of a scholarly source? And give an example of a scholarly source.
32. What is misinformation?
33. What is disinformation?
34. What is a peer reviewed source?
35. What is academic integrity?
36. What is plagiarism?
37. What is self-plagiarism?
38. What is the best way to avoid plagiarism?

Pt2520 Final Exam Study Guide

DJ Losen



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Unveiling the Magic of Words: A Overview of "**Pt2520 Final Exam Study Guide**"

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ve

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 5co i2 5co2 i 2 o 5 5 c o i 2 5 c o 2

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 reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation
 state and oxidation number concepts to identify redox reactions and to keep track of electrons

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 reducing agent also 2sr o2 2sro sr0 to sr2 oxidized reducing agent o0 to o2 reduced ox ag 2li s li2s li0 to li1 oxidized red ag
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