

KOYO BEARINGS

STEERING HEAD BEARING KIT CROSS REFERENCES

ORDER REF	BEARING	DUST	BEARING	DUST
	Upper	Seal	Lower	Seal
SH045	6205	N	30205	N
SH046	225H 41H	N	225H 41H	N
SSH050	225H 41H	N	24H 41H	BEA000/8
SSH200	225H 41H	N	24H 41H	BEA000/6
SSH500	H264815Y-26 48SH	N	27 48SH	N
SSH750	H264815Y-26 48SH	N	H305014Y	N
SSH901	32005JR	N	32006JR	BEA000/7
SSH902	91015-425-832	N	91015-425-832	BEA000
SSH902R	91015-425-832	BEA000	91015-425-832	BEA000/R
SSH902RR	91015-425-832	BEA000	91015-425-832	BEA000
SSH903	91015-425-832	N	32006JR	N
SSH903R	91015-425-832	BEA000	32006JR	BEA000/7
SSH904	91015-425-832	BEA000/5	CR-0627 (305115)	BEA000/1
SSH905	CR-0627 (305115)	BEA000/1	CR-0627 (305115)	BEA000/2
SSH907	CR-0627 (305115)	BEA000/2	32006JR	BEA000/7
SSH908	91015-425-832	BEA000/5	91015-425-832	BEA000
SSK100	30YM 48Y	N	30YM 48Y	N
SSK125	25M1 45M	N	284512	N
SSK200	25M1 45M	N	28YM 51M	N
SSK400	30K 48KS	N	30KW 52K	N
SSK901	320/28JR	N	32006JR	N
SSK902	32005JR	N	320/28JR	N
SSK904	32005JR	BEA000	32005JR	BEA000
SSK905	32006JR RS-2	N	320/28JR RS-2	Y
SSK906	32005JR	BEA000	320/28JR RS-2	Y
SSK907	32907JR-3	N	32907JR-3	N
SSK908	32005JR	N	32907JR-3	N
SSS100	25M1 45M	N	275 48KS	N
SSS250	255 48YS	N	275 48KS	N
SSS901	91015-425-832	N	32005JR	BEA000
SSS902	25YM 48Y	N	32006JR	N
SSS903	32006JR	N	32006JR	BEA000/7
SSS903R	32006JR	BEA000/7	32006JR	BEA000/7
SSS904	4T-CR-0643L	N	4T-CR-0643L	N
SSW055	34W 50W-HR 320/25	N	34W 50W-HR 320/25	N
SSW901	320/28JR	N	320/28JR	N
SSY080	25M1 43Y1	N	30YM1 48Y1	N
SSY085	25M1 43Y1	N	32006JR	N
SSY125	30YM 48Y	N	25YM 48Y	N
SSY500	25Y 48YS	N	SAC 3048	N
SSY901	320/28JR	N	320/28JR RS	Y
SSY902	320/28JR	N	320/28JR RS-1	Y

SPECIAL NONE CONVENTIONAL BEARINGS

Description	Dimensions	Order Ref
16005	47x25x8	BEA087
3205 (AFN9)	52x25x21	BEA089
3304 (AFN9)	52x20x22	BEA155
6204 SH4 - PC3	47x20x14	BEA492
6205 C4 PG Cage	52x25x19	BEA130
6304 SH2-9TC3	53x20x25	BEA404
6305R15H2-9TC535	52x25x16	BEA401
8KM354220-1UJ		
O.E. 93317-43580	40x34x18	BEA456
DG1742RS	42x17x18	BEA400
RLS5-LJ58"Y (STD)	40x16x11	BEA142
RLS6-LJ3/8"J (STD)	48x19x14	BEA143
RLS7-LJ38"J (STD)	51x22x14	BEA139
TAPER BEARINGS		
225H 41H	41x22 5x13	BEA440
24H 41H	41x24x12	BEA441
25M1 43Y1	43x25x11	BEA447
25M1 45M	45x25x14.5	BEA464
25M1 46M	46x25x12	BEA443
255 48YS	48x25x16	BEA428
25Y 48YS	48x25x15	BEA429
25YM 48Y	48x25x19	BEA430
27 48SH	48x27x15	BEA453
275 48KS	48x27x14	BEA431
284612	46x28x12	BEA474
28YM 51M	51x28x12	BEA444
30204JR	47x20x15	BEA159
30304JR	52x20x16	BEA183
30306JR	72x30x21	BEA182
30K 48KS	48x30x14	BEA463
30KW 52K	52x30x17	BEA445
30YM 48Y	48x50x13	BEA479
32004K	42x20x16	BEA483
320/28JR	52x28x16	BEA147
320/28JR RS	52x28x18	BEA154
320/28JR RS-1	52x28x17	BEA405
320/28JR RS-2	52x28x19	BEA478
32005JR	47x25x15	BEA106
32005JR RS	47x25x15	BEA144
32005JR RS-1	47x25x17	BEA160
32006JR	55x30x17	BEA105
32006JR RS	55x30x19	BEA129
32907 JR-2	55x35x15	BEA116
32907 JR-3	55x35x14	BEA126
34W 50W-HR 320/25	51x34x12	BEA446
4T-CR-0643L	52x30x16	BEA439
CR-0627 (305115)	51x30x15	BEA436
H264815Y-26 48SH	48x26x15	BEA432
H305014Y	50x30x15	BEA433
HONDA O.E.		
91015-425-832	47x26x15	BEA435



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Yijin Wang



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web in order to balance a chemical equation the quantities of each type of element and polyatomic ion that are present in the reactants and the products of the reaction must be determined because polyatomic ions contain multiple atoms and generally react as indivisible units any polyatomic ion that is present in both a reactant and a product in

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web a balanced chemical equation often may be derived from a qualitative description of some chemical reaction by a fairly simple approach known as balancing by inspection consider as an example the decomposition of water to yield molecular hydrogen and oxygen

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web oct 31 2023 to balance a chemical equation enter an equation of a chemical reaction and press the balance button the balanced equation will appear above use uppercase for the first character in the element and lowercase for the second character examples fe au co br c o n f ionic charges are not yet supported and will be ignored

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web balancing chemical equations is one of those concepts in chemistry that often confuses people but i think we ll see that if

we work through this carefully and methodically and we also appreciate the art of balancing chemical equations that it s actually not too bad

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web feb 19 2022 in a balanced chemical equation both the numbers of each type of atom and the total charge are the same on both sides equations ref 3 1 1 and ref 3 1 2 are balanced chemical equations what is different on each side of the equation is how the atoms are arranged to make molecules or ions

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web mar 14 2023 the ultimate goal for balancing chemical equations is to make both sides of the reaction the reactants and the products equal in the number of atoms per element this stems from the universal law of the conservation of mass which states that matter can neither be created nor destroyed

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web balancing equations the chemical equation described in section 4 1 is balanced meaning that equal numbers of atoms for each element involved in the reaction are represented on the reactant and product sides this is a requirement the equation must satisfy to be consistent with the law of conservation of matter

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web because of the law of conservation of mass balancing chemical equations gives a more accurate representation on what s happening in a reaction the accuracy of the ratios of moles is important in calculating the theoretical yield and how much stuff is reacting mass wise stoichiometry

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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide CaO and H_2O to form aqueous calcium hydroxide Ca(OH)_2 is it possible to balance the equation if you incorrectly identify the product as CaOH and if so what is the equation

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web balance a chemical equation when given the unbalanced equation explain the role of the law of conservation of mass in a chemical reaction even though chemical compounds are broken up and new compounds are formed during a chemical reaction atoms in the reactants do not disappear nor do new atoms appear to form the products

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web dec 2 2020 a balanced chemical equation tells you the amounts of reactants and products needed to satisfy the law of conservation of mass basically this means there are the same numbers of each type of atoms on the left side of the equation as there are on the right side of the equation

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web balance the following chemical equation $\text{Mg(OH)}_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$ note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem do 4 problems

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web chemists use symbols and formulae to represent elements and compounds word equations and balanced chemical equations represent the changes that happen in chemical reactions

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web oct 13 2023 method 1 doing a traditional balance download article 1 write down your given equation for this example you will use $\text{C}_3\text{H}_8 + 2\text{H}_2\text{O} \rightarrow \text{CO}_2$ this reaction occurs when propane C_3H_8 is burned in the presence of oxygen to produce water and carbon dioxide 2 write down the number of atoms per element

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