

Rock Candy Lab - Minerals from Solution

Name _____ Period _____ Date _____

Here's kitchen chemistry that turns a solution of sugar and water into a tasty treat.

Science is often referred to as the most fun subject in school mainly because... it *is* it can be a most delicious subject, too. When you make this treat, you'll be jumping "sick-first" into some serious chemistry that's hiding as candy. You'll discover and use principles like solutes, solvents, viscosity, sedimentation, solutions, and supersaturated solutions. How tasty does all that sound? Not so much? Read on!

Materials:

- Beaker 500mL or larger
- 2 $\frac{1}{2}$ c sugar
- $\frac{1}{4}$ c water (160 mL)
- Heavy spoon
- Food color
- Flavoring (ie: vanilla extract)
- Jar (large for a group of 2)
- Skinny Skewer (for each partner)
- Popsicle stick
- Post-it note
- Wax paper
- Paper towel + rubber band

Lab Instructions:

1. Pour 2 $\frac{1}{2}$ cups of granulated sugar into the beaker.
2. Add $\frac{3}{4}$ cup (160 mL) of water to the sugar. Watch what happens as the water bubbles through all that sugar. There's a lot going on in the container already. Use the heavy spoon to thoroughly stir **the water (a solvent)** and **the sugar (a solute)** together to make a solution. It will be very viscous (thick) and heavy because there's a lot more sugar than water in there. Stir it well!
3. You need to give the water some help with all that sugar so warm up the water. Put the solution on a hot plate, or in the microwave and heat it for 1 $\frac{1}{2}$ minutes on high. Heat the solution to the boiling point. **Don't let the solution boil over.** You may check to see if you can use another teacher's microwave if there is a line at our microwave (take a towel with you for when you touch the hot beaker).
4. **CAUTION:** Use oven mitts to handle the hot solution and move it to a stable, heat-safe location. Use the heavy spoon to thoroughly stir the solution again. Make sure all the sugar is stirred but watch out for splatters of hot liquid. Notice how the solution is



changing by just using heat. Describe the changes you see happening:

-
5. Move the stirred solution to the hot plate or microwave again and heat it on high for another minute. **Don't let the solution boil over.**

Recipe For Rfast Rock Candy Lab

Gardner Dexter Hiscox



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web apr 18 2011 the aspirin will be titrated against a standard solution of base 0.100 M NaOH base will be dispensed from a burette into a beaker containing the dissolved in ethanol acid and phenolphthalein indicator which

aspirin screen experiment resource rsc education - Jun 04 2023

web the aspirin screen experiment has been designed to be a flexible open tool for teachers and students which allows students to run their own reaction online before taking part in the real thing this also features an in depth practical guide and a set of accompanying worksheets covering the theory

written report of practical activity aspirin book waptac - Jul 05 2023

web written by a leading expert on aspirin related research this is the most comprehensive treaty of the history pharmacological effects and clinical applications of one of the most successful drugs ever

[preparation of aspirin sserc](#) - Dec 30 2022

web within this section you can find a large selection of teaching resources covering a range of science and technology areas the vast majority of these involve hands on practical work and range from simple experiments for early years through to advanced higher project ideas and teacher demonstrations

[aspirin as one of the major and oldest pain killers studentshare](#) - Jun 23 2022

web aspirin aspirin aspirin is one of the major and oldest pain killers used in the medical science it is assumed that almost 40000 tonnes of aspirin are consumed every year worldwide it has many other medical uses in the treatment of fever inflammation swelling rheumatoid arthritis rheumatic fever and mild infection