#### Introduction to Solubility PhET Lab (council 3/2011)

We can express concentrations in terms of moles solute per liters of solution

#### Introduction

When we cook pasta, we frequently dissolve table salt in water. Is there a limit to how much salt we can get into the water?

When ionic-bonded compounds (like table salt, NaCl) are dissolved in polar solvents (like water), the **polar** water molecules act to pull the ions in the compound apart. When this happens (called dissociation), ions are released into the solution. Is there a limit to the number of ions that can exist in solution?



Salts & Solubility

#### Pre-lab Vocabulary

Please use	NOME	terethoods.	NUMBER	more services.	4000	the	internet.	505	define.	(ber	following terms:
A DESCRIPTION OF THE PERSON	A Comment	THE RESERVE AND ADDRESS OF THE PARTY OF THE	2 10 10 10 10 10 10 10 10 10 10 10 10 10	THE RESERVE OF THE PARTY OF THE	100,000	The second second	THE RESIDENCE OF THE PARTY OF	10000	Company of the Control of the	10.00	THE SECOND PROPERTY AND ADDRESS OF THE PARTY

Saturated	
Unsaturated	
Supersaturated	
Molarity .	
Cation	
Supersaturated Molarity Cation Anion Dissociation	
Dissociation	
Solubility	

Procedure PhET Simulations → Play With Sims → Chemistry → Salts and Solubility →

Table Soll. Take some time and familiarize yourself with the simulation before answering the below questions:

- What happens to the bound Na+ and CI- ions as they enter the water?
- 2. Is there a point at which no more sodium and chloride ions will dissolve? How do you know?
- 3. What do we call a solution that has as many ions as it can hold?
- 4. What happens if you add additional, solid NaCl after the maximum has been reached?
- 5. What happens when more water is added to a saturated solution?
- 6. What happens to the ions bound together as a solid when the solution's volume is reduced?
- 7. How many total Na\* ions can you add until no more will dissolve (the maximum amount)?
- How many moles does this many ions represent? (recall, 1 mole = 6.02 x 10<sup>23</sup> things)
- What volume (in Liters) did you use to get the saturation point in #7 above?
- What concentration is this? (In mole/L, use #8 and #9's answers) This is molarity, M.

# **Phet Into To Solubility Simulations Answer Key**

**S Ben Porath** 

**Phet Into To Solubility Simulations Answer Key:** 

#### Decoding Phet Into To Solubility Simulations Answer Key: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Phet Into To Solubility Simulations Answer Key**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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