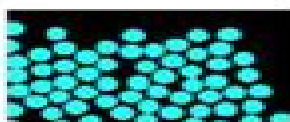


## PhET Interactive Simulation: States of Matter

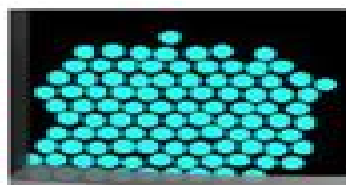
[Click Here for the States of Matter Simulation](#), practice then answer the questions below:  
(Be sure to download the simulation before starting).

### Exploration

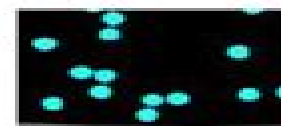
1. Identify the following as either a solid, liquid, or a gas.



a) liquid



b) solid



c) gas

2. a) What happens to the atoms of a solid when the temperature increases?  
b) decreases? *When the temp increases in a solid the atoms start to spread out, when the temp decreases the atoms get more and more compact.*
3. What happens to the atoms of a solid when the temperature decreases?  
*The atoms get more and more compact.*
4. Complete the following table for the initial temperature of each substance in each state:

	Solid	Liquid	Gas
Neon	-295 c	-246 c	-217 c
Argon	-230 c	-187 c	-84 c
Oxygen	-246 c	-216 c	-160 c
Water	-127 c	13 c	156 c

5. Use the table above to compare any two (2) of the gases. How do their initial temperatures compare at each physical state?  
*When neon gets very cold it becomes a gas as you can see from the chart, but when water gets very hot it becomes a gas.*
6. a) What happens to the pressure if the temperature is increased?  
b) decreased? *When the temp is increased the pressure goes up for everyone of them except oxygen and when the temp is decreased the pressure goes down for everyone except oxygen. For oxygen when the temp is increased pressure goes down while decreasing it makes the pressure go up.*

# Phet Gases Simulation Answers

**CL Gary**



## **Phet Gases Simulation Answers:**

## Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Phet Gases Simulation Answers**

In a world inundated with screens and the cacophony of immediate connection, the profound energy and emotional resonance of verbal art frequently fade in to obscurity, eclipsed by the continuous onslaught of noise and distractions. However, nestled within the lyrical pages of **Phet Gases Simulation Answers**, a fascinating function of fictional beauty that impulses with natural feelings, lies an remarkable journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, this enchanting opus guides readers on a mental odyssey, lightly revealing the latent possible and profound influence stuck within the intricate internet of language. Within the heart-wrenching expanse of this evocative examination, we can embark upon an introspective exploration of the book is central subjects, dissect their fascinating publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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