| Name:                         | Date:                                                                                 | Date:                  |  |
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| PHASE C                       | HANGE WOR                                                                             | KSHEET                 |  |
| On the line at the left,      | write the term that best matches e                                                    | ach description below: |  |
| condensation                  | vaporization                                                                          | melting point          |  |
| phase change                  | melting                                                                               | sublimation            |  |
| volatile                      | equilibrium vapor pressure                                                            | deposition             |  |
| sublimation                   | _ 1) conversion of solid directly into a gas                                          |                        |  |
| condensation                  | _ 2) opposite of vaporization                                                         |                        |  |
| melting point                 | 3) temperature at which solid and liquid                                              |                        |  |
|                               | phases exist in equilibrium.                                                          |                        |  |
| phase change                  | 4) conversion of a substance from one                                                 |                        |  |
|                               | of the three states of matter to another                                              |                        |  |
| vaporization                  | _ 5) change from a liquid to gas                                                      |                        |  |
| equilibrium vapor<br>pressure | 6)                                                                                    |                        |  |
|                               | _ 6) pressure exerted by a constant number of gas molecules in equilibrium with their |                        |  |
|                               | liquid phase                                                                          | briam with their       |  |
| volatile                      | _ 7) description of a liquid tha                                                      | t evanorates           |  |
|                               | easily, with a low boiling p<br>vapor pressure                                        |                        |  |
| deposition                    | _ 8) transformation of a gas o                                                        | directly into a solid  |  |
| melting                       | _ 9) phase change from a solid to a liquid                                            |                        |  |

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