CRITICAL POINT - The temperature above which a substance will always be a gas regardless of the pressure.

Pressure (atm)

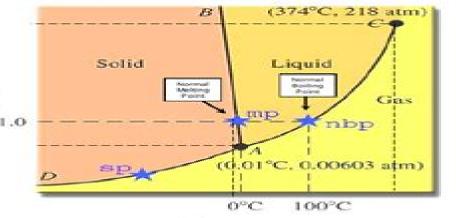
- NOTE:
  - The solid phase is more dense than the liquid phase.
  - The line between the solid and gas phases is the equilibrium of solid and gas phases at that specific pressure and temperature, i.e. a curve of all the deposition/sublimation points.
  - The line between the solid and liquid phases is the equilibrium of solid and liquid phases at that specific pressure and temperature, i.e. a curve of all the freezing/melting points.
  - The line between the liquid and gas phases is the equilibrium of liquid and gas phases at that specific pressure and temperature, i.e. a curve of all the vaporization/condensation points.
- Melting Point (Freezing Point) The temperature at which the solid and liquid phases of a substance are in equilibrium at atmospheric pressure.
  - Normal Melting Point (Freezing Point) The temperature at which the solid changes to a liquid at Standard Pressure (1.00 atm = 760 mmHg = 760 torr = 101.325 kPa)
- Boiling Point (Condensation Point) The temperature at which the vapor pressure of a liquid is equal
  to the pressure on the liquid.
  - Normal Boiling Point (Condensation Point) The temperature at which the vapor pressure of a liquid is equal to Standard Pressure (1.00 atm = 760 mmHg = 760 torr = 101.325 kPa)

## Phase Diagram for Water

For water, the liquid phase is more dense than the solid phase due to hydrogen bonding.

liquid, and gas phases exist

simultaneously.



Temperature

# **Phase Change Diagram Ws Answer**

John W. Goodby,Peter J.
Collings,Takashi Kato,Carsten
Tschierske,Helen Gleeson,Peter
Raynes,Volkmar Vill

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source is intermittent such as Solar Energy The use of intermittent energy sources is likely to grow If more and more solar energy is to be used for domestic and industrial applications then energy storage is very crucial If no storage is used in solar energy systems then the major part of the energy demand will be met by the back up or auxiliary energy and therefore the so called annual solar load fract on will be very low In case of solar energy both short term and long term energy storage systems can be used which can adjust the phase difference between solar energy supply and energy demand and can match seasonal demands to the solar availability respectively Thermal energy storage can lead to capital cost savings fuel savings and fuel substitution in many application areas Developing an optimum thermal storage system is as important an area of research as developing an alternative source of energy Thermodynamics and Kinetics in Materials Science Boris S. Bokstein, Mikhail I. Mendelev, David J. Srolovitz, 2005-06-30 Accompanying CD ROM contains computer tests and laboratories CD ROM label Position, Navigation, and Timing Technologies in the 21st Century Y. Jade Morton, Frank van Diggelen, James J. Spilker, Jr., Bradford W. 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papers presented were carefully reviewed and selected from 320 submissions. In addition the book contains the abstracts of 2 keynote speeches. The contributions are organized in topical sections on requirements adaptation and evolution model transformation conceptual design domain specific languages case studies and experiences mining and matching business process modelling validation and quality and service and management. A Treatise on the Theory of Solution,

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Anne Conzemius, Jan O'Neill, 2009-04-01. Help staff focus on results and implement SMART Strategic and specific Measurable. Attainable Results based and Time bound goals to transform your school into a place where every student meets or exceeds standards. The authors present four success stories from real SMART schools and several frameworks for adult and student goal setting that lead to real results. Dimensions, 1955. Technical News Bulletin, 1956. Technical News Bulletin.

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