

## pH Practice Worksheet

- 1) What is the pH of a solution that contains 25 grams of hydrochloric acid (HCl) dissolved in 1.5 liters of water?
  
  
  
  
  
  
  
  
  
  
- 2) What is the pH of a solution that contains 1.32 grams of nitric acid (HNO<sub>3</sub>) dissolved in 750 mL of water?
  
  
  
  
  
  
  
  
  
  
- 3) What is the pH of a solution that contains 1.2 moles of nitric acid (HNO<sub>3</sub>) and 1.7 moles of hydrochloric acid (HCl) dissolved in 1000 liters of water?
  
  
  
  
  
  
  
  
  
  
- 4) If a solution has a [H<sup>+</sup>] concentration of  $4.5 \times 10^{-7}$  M, is this an acidic or basic solution? Explain.
  
  
  
  
  
  
  
  
  
  
- 5) An acidic solution has a pH of 4. If I dilute 10 mL of this solution to a final volume of 1000 mL, what is the pH of the resulting solution?

# Ph Review Problems Chemfiesta Answers

**DP Hallahan**



## **Ph Review Problems Chemfiesta Answers:**

## Unveiling the Magic of Words: A Review of "**Ph Review Problems Chemfiesta Answers**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Ph Review Problems Chemfiesta Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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