# Lab: PhET Sugar and Salt Solutions

Name: Date: Class:

This lab is a computer-simulation-based activity. The simulation requires Java and may not run on all computers. If it won't run on student machines the activity can be done by showing the simulator using a computer-projection system. The simulator is available for download at https://phet.colorado.edu/en/simulation/legacy/sugar-and-salt-solutions. Or, perhaps more easily accessed by performing a search for "PhET Sugar and Salt".

# Background

Sugar and salt both dissolve easily in water and they look a lot alike. There is an important difference between them, though. Salt is an ionic compound made up of two kinds of ion: sodium ions (Na<sup>+</sup>) and chloride ions (Cl<sup>-</sup>). Sugar is a molecular compound made of one kind of molecule: C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>.

Though it looks the same to the naked eye when you add a spoonful of either of these substances to water, what happens to the ions or molecules is quite different. The ions of the salt separate and become surrounded by water molecules. The chemical formula of salt is broken in half and instead of having just one kind of thing dissolved in the water there are two, a positive ion and a negative ion. See the image at right.

When sugar dissolves the molecules separate from each other but the chemical formula stays together. There is just one kind of thing dissolved in the water and it is neutrally charged. See the image at right.

Ionic compounds are also called **electrolytes**. An electrolyte is a material which breaks up into ions when it dissolves in water. When electrolytes dissolve in water they cause the water to be able to conduct electricity. Molecular compounds are usually **non-electrolytes**. A non-electrolyte is a material that dissolves in water as whole molecules: no ions are produced and the water does not gain the ability to conduct electricity.

# Questions

Put the conductivity tester into the water in the simulator on the "Macro" tab.

1. What happens when you add salt to the water? How does it change as you add more and more salt?

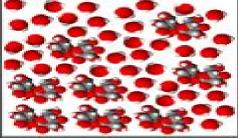
The light bulb illuminates, with less the light is dim

with more the light is bright.

What does "concentration" mean? Refer to the graph in the upper-right corner of the sim.

concentration is the amount of solute in a mixture

3. Reset All using the button in the lower-right corner. What happens when you add sugar to the water? How does it change as you add more sugar? Salt dissolving in water. The salt ions are positive and negative. The water molecules have one large oxygen atom and two small hydrogen atoms.



Sugar dissolving in water. The sugar molecules are the large groups of aroms. The molecules in the solid are stack sugether but the dissolved sugar molecules move separately while remaining whole sugar molecules.

there is no change. the light stays off entire time. does not change with amount of solute

4. Reset All again. Add all of the salt. What changes when you drain water out? Explain.

the light remains on and un changing.

Reset All again. Add all of the salt. What changes when you add water? Explain.

nothing changes. light stays off

Reset All again. Add all of the salt. What changes when you evaporate water out? Explain.

the light gets brighter

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# **Table of Contents Ph Of Salt Solutions Physical Science**

- 1. Understanding the eBook Ph Of Salt Solutions Physical Science
  - The Rise of Digital Reading Ph Of Salt Solutions Physical Science
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Ph Of Salt Solutions Physical Science
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ph Of Salt Solutions Physical Science
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Ph Of Salt Solutions Physical Science
  - Personalized Recommendations
  - $\circ\,$  Ph Of Salt Solutions Physical Science User Reviews and Ratings
  - Ph Of Salt Solutions Physical Science and Bestseller Lists
- 5. Accessing Ph Of Salt Solutions Physical Science Free and Paid eBooks
  - Ph Of Salt Solutions Physical Science Public Domain eBooks
  - Ph Of Salt Solutions Physical Science eBook Subscription Services
  - Ph Of Salt Solutions Physical Science Budget-Friendly Options

- 6. Navigating Ph Of Salt Solutions Physical Science eBook Formats
  - o ePub, PDF, MOBI, and More
  - Ph Of Salt Solutions Physical Science Compatibility with Devices
  - Ph Of Salt Solutions Physical Science Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Ph Of Salt Solutions Physical Science
  - Highlighting and Note-Taking Ph Of Salt Solutions Physical Science
  - Interactive Elements Ph Of Salt Solutions Physical Science
- 8. Staying Engaged with Ph Of Salt Solutions Physical Science
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ph Of Salt Solutions Physical Science
- 9. Balancing eBooks and Physical Books Ph Of Salt Solutions Physical Science
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ph Of Salt Solutions Physical Science
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Ph Of Salt Solutions Physical Science
  - Setting Reading Goals Ph Of Salt Solutions Physical Science
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ph Of Salt Solutions Physical Science
  - Fact-Checking eBook Content of Ph Of Salt Solutions Physical Science
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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