



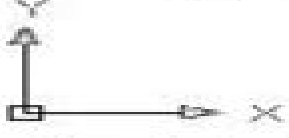

## Student Directions for *Gas Properties* Chemistry: Gas Laws

<http://phet.colorado.edu/>

### Learning Goals:

- Design experiments to measure the relationships between pressure, volume, and temperature.
- Create graphs based on predictions and observations.
- Make qualitative statements about the relationships between pressure, volume and temperature using molecular models.

**Predictions:** Make a chart like the one below. Without using the simulation, sketch what you think the graphs would look like. **Note: Be sure to label your x and y axes.**

I. Volume-Pressure graph 	Explain your reasoning for the graph's appearance
II. Volume-Temperature graph 	Explain your reasoning for the graph's appearance
III. Temperature-Pressure graph 	Explain your reasoning for the graph's appearance
IV. Number of particles – Volume 	Explain your reasoning for the graph's appearance

### Experiments:

1. For each case, I-IV. Write a short description of how to use the sim to collect data. Then make an Excel spreadsheet for each, graph and curve fit. Some helpful hints – if you set a parameter like temperature constant, then make a change, you

# Phet Gas Properties Lab Answers

**C. Carey**



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