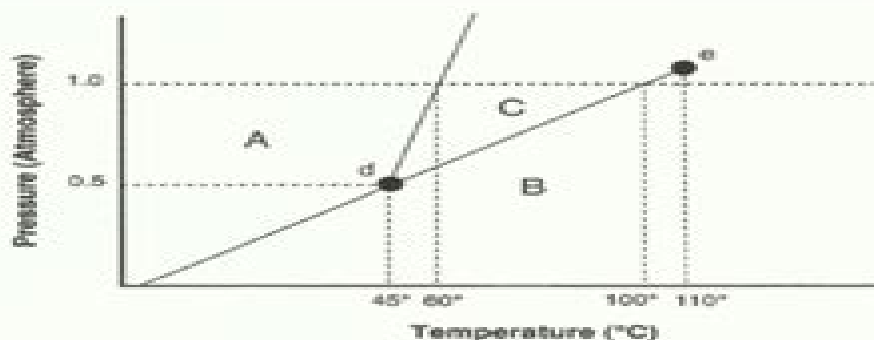


KEY

PHASE DIAGRAM WORKSHEET

Part A – Generic Phase Diagram.

Answer the questions below in relation to the following generic phase diagram.



1. Which section represents the solid phase? **A**
2. What section represents the liquid phase? **C**
3. What section represents the gas phase? **B**
4. What letter represents the triple point? **d**

In your own words, what is the definition of a triple point?

It is the point where all three phases coexist at a specific temperature and pressure

5. What is this substance's normal melting point, at 1 atmosphere of pressure? **60°C**
6. What is this substance's normal boiling point, at 1 atmosphere of pressure? **100°C**
7. Above what temperature is it impossible to liquefy this substance, no matter what the pressure?

110°C (This is known as the critical temperature)

8. At what temperature and pressure do all three phases coexist? **45°C and 0.5 atm**
9. At a constant temperature, what would you do to cause this substance to change from the liquid phase to the solid phase? **You would need to increase the pressure.**
10. What does sublimation mean? **The process of going from a solid to a gas directly, skipping the liquid phase altogether.**

Phase Diagram Answer Key

**Hugh D Young, Roger A
Freedman, Ragbir Bhathal**



Phase Diagram Answer Key:

Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation Pharmaceutics Philip Denton, Chris Rostron, 2013-07-25 Pharmaceutics the science of medicine design explores the different forms that medicines can take and demonstrates how being able to select the best form be it a tablet injectable liquid or an inhaled gas requires an understanding of how chemicals behave in different physical states , **UPSC Geo-Scientist Pre Hydrology/Geology Question Bank Book 1800+ MCQ With Detail Solution** DIWAKAR EDUCATION HUB , 2024-09-24 UPSC Geo Scientist Pre Hydrology Geology Question Bank Book 1800 MCQ With Detail Solution Highlight of Book Topic Wise MCQ with Detail Solution Design by Expert Faculty As Per New Updated Syllabus As Per UPSC Geo Scientist Prelims Syllabus **Student Solutions Manual for Physical Chemistry** C. A. Trapp, Peter Atkins, Julio dePaula, 2009-12-18 With its modern emphasis on the molecular view of physical chemistry its wealth of contemporary applications vivid full color presentation and dynamic new media tools the thoroughly revised new edition is again the most modern most effective full length textbook available for the physical chemistry classroom Available in Split Volumes For maximum flexibility in your physical chemistry course this text is now offered as a traditional text or in two volumes Volume 1 Thermodynamics and Kinetics ISBN 1 4292 3127 0 Volume 2 Quantum Chemistry Spectroscopy and Statistical Thermodynamics ISBN 1 4292 3126 2 *SAT Subject Test Chemistry* Joseph A. Mascetta, Mark Kernion, 2020-12-01 The updated edition of Barron's SAT Subject Test Chemistry includes A full length diagnostic test with explained answers Four practice tests that reflect the actual SAT Subject Test Chemistry All questions answered and explained Detailed reviews covering all test topics Appendixes which include the Periodic Table important equation constant and data tables and a glossary of chemistry terms Both teachers and test taking students have praised earlier editions of this manual for its wealth of well organized detail Subject reviewed include the basics matter energy scientific method and measurements atomic structure and the periodic table bonding chemical formulas gases and laws stoichiometry liquids solids and phase changes chemical reactions and thermochemistry chemical reactions chemical equilibrium acids bases and salts oxidation reduction carbon and organic chemistry and the laboratory ONLINE PRACTICE

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Chemistry, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

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Hugh D Young, Roger A Freedman, Ragbir Bhathal, 2010-08-04 This book is the product of more than half a century of leadership and innovation in physics education When the first edition of University Physics by Francis W Sears and Mark W Zemansky was published in 1949 it was revolutionary among calculus based physics textbooks in its emphasis on the fundamental principles of physics and how to apply them The success of University Physics with generations of several million students and educators around the world is a testament to the merits of this approach and to the many innovations it has introduced subsequently In preparing this First Australian SI edition our aim was to create a text that is the future of Physics Education in Australia We have further enhanced and developed University Physics to assimilate the best ideas from education research with enhanced problem solving instruction pioneering visual and conceptual pedagogy the first systematically enhanced problems and the most pedagogically proven and widely used online homework and tutorial system

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and talkative with his family his delirium seems to be slowly improving blood pressure remains a little bit elevated still

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