

Stoichiometry Review

1. The combustion of propane occurs via the reaction:



How many grams of oxygen are required to burn completely 10.0 g of propane?

$$\left(\frac{10.0 \text{ g C}_3\text{H}_8}{44.09652 \text{ g/mol}} \right) \left(\frac{5 \text{ mol O}_2}{1 \text{ mol C}_3\text{H}_8} \right) \left(31.9988 \text{ g/mol} \right) = 36.3 \text{ g O}_2$$

2. Small quantities of chlorine can be prepared in the laboratory by the reaction:



What volume of chlorine at STP can be prepared from 100.0 g of manganese dioxide?

$$\left(\frac{100.0 \text{ g MnO}_2}{86.9368 \text{ g/mol}} \right) \left(\frac{1 \text{ mol Cl}_2}{1 \text{ mol MnO}_2} \right) \left(22.4 \text{ L/mol} \right) = 25.8 \text{ L Cl}_2$$

3. The most common ore of arsenic is mispickel, FeSAs. Upon heating this ore, free arsenic is obtained. How many grams of FeSAs are required to produce 10.0 g of arsenic according to the following reaction?



$$\left(\frac{10.0 \text{ g As}}{74.9216 \text{ g/mol}} \right) \left(\frac{1 \text{ mol FeSAs}}{1 \text{ mol As}} \right) \left(162.8346 \text{ g/mol} \right) = 21.7 \text{ g FeSAs}$$

Pg 31section Review Stoichiometry Answers

Karin Nielsen-Saines



Pg 31section Review Stoichiometry Answers:

The Enigmatic Realm of **Pg 31section Review Stoichiometry Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Pg 31section Review Stoichiometry Answers** a literary masterpiece penned by a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of people who partake in its reading experience.

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