

Name: _____ per: _____

Worksheet - Reaction Rates

Use this reaction for the questions below: $C_6H_{12}O_6 (s) + 6 O_2(g) \rightarrow 6 H_2O (g) + 6 CO_2 (g)$

- What happens to the concentrations of:
 - $C_6H_{12}O_6$ & O_2 as the reaction proceeds $\rightarrow$?
 - H_2O + CO_2 as the reaction proceeds $\rightarrow$?
- According to the collision theory, what 3 circumstances are needed for $C_6H_{12}O_6$ & O_2 to react?
- What is the activation energy for a chemical reaction?
- Use the equation & the collision theory to explain:



Change in condition:	Does this <u>increase</u> or <u>decrease</u> the rate of reaction?	Explain why
a. <u>Increasing the temperature</u>	Ex: Increase (springs up)	Ex: Molecules move faster & collide more = increased rxn rate
b. <u>Increasing the concentration</u> of $C_6H_{12}O_6$		
c. <u>Decreasing the concentration</u> of O_2		
d. <u>Increase the surface area</u> by chewing up food in your mouth.		
e. <u>Decreasing the temperature</u>		
f. <u>Increasing the pressure</u> in the container		
g. <u>Decreasing the concentration</u> of H_2O		
h. <u>Increasing the volume</u> of the container the reaction occurs in		
i. <u>Increasing the concentration</u> of CO_2		
j. <u>Using a catalyst</u> (like salivary amylase)		

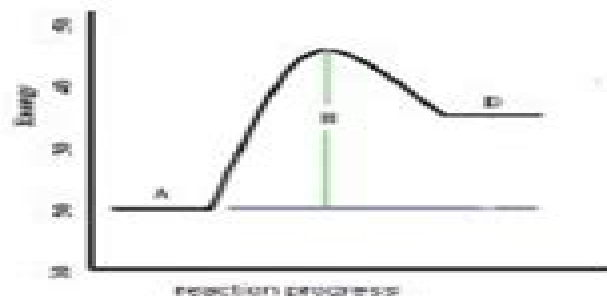
- On the accompanying energy diagram, label the following terms:

a. reactants b. products c. activation energy

- On the graph to the right, draw and label what this diagram would look like if a catalyst was added to the reaction.

- Graph reading

- How much energy (kJ) do the reactants have?
- How much energy (kJ) do the products have?
- How much energy (kJ) is required to activate this un-catalyzed reaction?
- Is this reaction endothermic or exothermic? How do you know?



Reaction Rates Equilibrium Worksheet Answers Chapter 19

Judd E. Hollander



Reaction Rates Equilibrium Worksheet Answers Chapter 19:

Unveiling the Energy of Verbal Artistry: An Mental Sojourn through **Reaction Rates Equilibrium Worksheet Answers Chapter 19**

In a global inundated with monitors and the cacophony of fast interaction, the profound energy and psychological resonance of verbal artistry often diminish into obscurity, eclipsed by the regular barrage of noise and distractions. However, nestled within the musical pages of **Reaction Rates Equilibrium Worksheet Answers Chapter 19**, a captivating function of literary beauty that impulses with fresh feelings, lies an unforgettable journey waiting to be embarked upon. Composed by way of a virtuoso wordsmith, this mesmerizing opus instructions viewers on a psychological odyssey, lightly revealing the latent possible and profound affect stuck within the intricate web of language. Within the heart-wrenching expanse with this evocative evaluation, we shall embark upon an introspective exploration of the book is main themes, dissect its captivating publishing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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