#### Types of Chemical Reactions

Name\_\_\_\_

Period\_\_\_\_\_Date\_\_\_\_

<u>Directions:</u> Label the following reactions as synthesis (S), decomposition (D), single replacement (SR), double replacement (DR) and combustion (C), then balance. Place a "B" at the end that are balanced as written.

3. 
$$C_2H_4 + O_2 \rightarrow CO_2 + H_2O$$

4. 
$$H_2O \rightarrow H_2 + O_2$$

5. 
$$H_2 + O_2 \rightarrow H_2O$$
 \_\_\_\_\_

6. 
$$C_8H_{18} + O_2 \rightarrow CO_2 + H_2O$$

11. 
$$C_6H_{12} + O_2 \rightarrow CO_2 + H_2O$$

18. 
$$C_6H_8 + O_2 \rightarrow CO_2 + H_2O$$
 \_\_\_\_\_

20. 
$$H_2 + N_2 \rightarrow NH_3$$

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