

		gerangskik ☒ is om die hoeke tussen hulle te maksimeer. ☒ Strukturele elektronpare is bindingspare plus alleenpare. OF Valensdop-elektronpaarafstotingsmodel is 'n model om die vorms van molekules te voorspel ☒ waarin die elektronpaar vanaf die buitenste dop ☒ van 'n verwysingsatoom rondom hierdie atoom gerangskik is om die afstoting tussen hulle te minimaliseer. ☒ 3 marks/punte	
	1.2.4	The molecule is a tetrahedron ☒ Die molekule is 'n tetraëder ☒ 1 mark/punt	LO2
	1.2.5	 Name trigonal pyramid ☒ Noem trigonale piramidele ☒ 3 marks/punte	LO4
1.3	1.3.1	The electron pair is equally shared between the two Cl atoms ☒ and the resulting Cl₂ molecule has no degree of polarity. ☒ Die elektronpaar word ewe veel gedeel tussen die twee Cl-atome ☒ en die resulterende Cl₂-molekule het geen mate van polariteit nie ☒. 2 marks/punte	LO2
	1.3.2	The tendency ☒ of an atom to pull the bonded pair ☒ of electrons towards itself. ☒ Die neiging ☒ van 'n atoom om die gebinde elektronpare ☒ na homself te trek ☒. 3 marks/punte	LO2
	1.3.3	Electronegativity of Na = 6 Electronegativity of Cl = 3 Difference is 3 ☒ With a difference of greater than 2.1 ☒ electron transfer will take place and the bond ☒ is ionic. ☒ (value of 2.1 does not need to be given for this mark) Elektronegatiwiteit van Na = 6 Elektronegatiwiteit van Cl = 3 Verskil is 3 ☒ Met 'n verskil van meer as 2.1 ☒ sal elektroneoordrag plaasvind en die binding is ionies. ☒ (waarde van 2.1 hoef nie vir hierdie punt gegee te word nie) 4 marks/punte	LO2

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web and now now endon will be king jarred thought he shook his head trying to make himself believe it he and endon had been friends since they were young children but what a difference there was between them for endon was the son of the king and queen the prince of deltora