

Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

9.01

B

Boron

12.01

C

Carbon

14.01

N

Nitrogen

16.00

O

Oxygen

19.00

F

Fluorine

20.18

Ne

Neon

22.99

Na

Sodium

24.31

Mg

Magnesium

26.98

Al

Aluminum

28.09

Si

Silicon

30.97

P

Phosphorus

32.07

S

Sulfur

35.45

Cl

Chlorine

39.96

Ar

Argon

39.10

K

Potassium

40.08

Ca

Calcium

44.96

Sc

Scandium

47.88

Ti

Titanium

50.94

V

Vanadium

52.00

Cr

Chromium

54.94

Mn

Manganese

55.85

Fe

Iron

58.93

Co

Cobalt

58.93

Ni

Nickel

63.55

Cu

Copper

65.39

Zn

Zinc

69.72

Ga

Gallium

72.64

Ge

Germanium

74.92

As

Arsenic

78.96

Se

Selenium

79.90

Br

Bromine

83.80

Kr

Krypton

85.47

Rb

Rubidium

87.62

Sr

Strontium

88.91

Y

Yttrium

91.22

Zr

Zirconium

92.91

Nb

Niobium

95.94

Mo

Molybdenum

98.91

Tc

Technetium

101.07

Ru

Ruthenium

102.91

Rh

Rhodium

106.42

Pd

Palladium

107.87

Ag

Silver

112.41

Cd

Cadmium

114.82

In

Indium

118.71

Sn

Tin

121.76

Sb

Antimony

127.60

Te

Tellurium

126.91

I

Iodine

131.29

Xe

Xenon

132.91

Cs

Cesium

137.33

Ba

Barium

138.91

La

Lanthanum

175.07

Hf

Hafnium

180.95

Ta

Tantalum

183.84

W

Tungsten

186.21

Re

Rhenium

188.91

Os

Osmium

190.23

Ir

Iridium

195.08

Pt

Platinum

196.97

Au

Gold

200.59

Hg

Mercury

204.38

Tl

Thallium

207.2

Pb

Lead

208.98

Bi

Bismuth

209

Po

Polonium

210

At

Astatine

210

Rn

Radon

223

Fr

Francium

226

Ra

Radium

227

Ac

Actinium

232

Rf

Rutherfordium

233

Ha

Hassium

238

Sg

Seaborgium

244

Bh

Bohrium

247

Hs

Hassium

251

Mt

Mendelevium

atomic number

11

28.09

atomic weight

symbol:

Si

Silicon

black

solid

white

synthetically prepared

near ambient temp

blue

liquid

red

gas

alkali metals

alkaline earth metals

transitional metals

other metals

nonmetals

noble gases

LABS

5

6

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10

11

12

13

14

15

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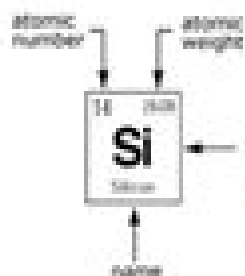
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symbol
 black solid
 blue liquid
 red gas
 white synthetically prepared
 over water layer

- alkali metals
- alkaline earth metals
- transitional metals
- other metals
- nonmetals
- noble gases

58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium

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Hume's Natural Philosophy and Philosophy of Physical Science Matias Slavov,2020-10-15 This book contextualizes David Hume s philosophy of physical science exploring both Hume s background in the history of early modern natural philosophy and its subsequent impact on the scientific tradition Drawing on Cartesian cosmology and Einstein s special relativity and taking in topics including experimentalism

causation laws of nature metaphysics of forces mathematics relation to nature and the concepts of space and time this book deepens our understanding of Hume's relation to natural philosophy It does so in addition by situating Hume's thought within the context of other major philosophers and scientists including Descartes Locke Boyle Kant Newton and Leibniz Demonstrating above all Hume's understanding of the fluid relationship between philosophy and science Hume's *Natural Philosophy* and *Philosophy of Physical Science* will provide new insights for historians and philosophers of science

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Physical Science Royal Society (Great Britain), 1911 *The Philosophy of Physical Science* Sir Arthur Stanley Eddington, 1939 *Peterson's Grad Programs in Physical Sciences, Math, Ag Sciences, Envir & Natural Res 20154 (Grad 4)* Peterson's, 2014-10-21 Graduate Programs in the Physical Sciences Mathematics Agricultural Sciences the Environment Natural Resources 2015 contains more than 3 000 graduate programs in the relevant disciplines including agriculture and food sciences astronomy and astrophysics chemistry physics mathematics environmental sciences and management natural resources marine sciences and more Informative data profiles for more than 3 000 graduate programs at nearly 600 institutions are included complete with facts and figures on accreditation degree requirements application deadlines and contact information financial support faculty and student body profiles Two page in depth descriptions written by featured institutions offer complete details on specific graduate programs schools or departments as well as information on faculty research Comprehensive directories list programs in this volume as well as others in the graduate series **Chemistry**

Calculations for Beginners John Obimakinde, Samuel Obimakinde, Ebenezer Obimakinde, Fredrick Akinbolade, 2025-05-30 With decades of combined experience as science teachers at both school and undergraduate levels the authors have recognised that one of the greatest challenges faced by students studying chemistry is grasping the complexity of the numerous numerical problems found in most parts of the subject This text is crafted to provide a clear and accessible pathway to overcoming this challenge by assisting students especially novices or those with minimal knowledge of the subject in performing chemistry calculations The content covers fundamental calculations crucial to understanding the principles of chemistry making it an invaluable tool for students aiming to excel in their studies Key features Designed with a

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how to understand the meaning of research terms how to disseminate research in a style which is clear and easily
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