

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

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L Darling-Hammond



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Table of Contents Reactions Study Guide Answers

1. Understanding the eBook Reactions Study Guide Answers
 - The Rise of Digital Reading Reactions Study Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Reactions Study Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reactions Study Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Reactions Study Guide Answers
 - Personalized Recommendations
 - Reactions Study Guide Answers User Reviews and Ratings
 - Reactions Study Guide Answers and Bestseller Lists
5. Accessing Reactions Study Guide Answers Free and Paid eBooks
 - Reactions Study Guide Answers Public Domain eBooks
 - Reactions Study Guide Answers eBook Subscription Services
 - Reactions Study Guide Answers Budget-Friendly Options

6. Navigating Reactions Study Guide Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Reactions Study Guide Answers Compatibility with Devices
 - Reactions Study Guide Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reactions Study Guide Answers
 - Highlighting and Note-Taking Reactions Study Guide Answers
 - Interactive Elements Reactions Study Guide Answers
8. Staying Engaged with Reactions Study Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reactions Study Guide Answers
9. Balancing eBooks and Physical Books Reactions Study Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reactions Study Guide Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Reactions Study Guide Answers
 - Setting Reading Goals Reactions Study Guide Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Reactions Study Guide Answers
 - Fact-Checking eBook Content of Reactions Study Guide Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

-
- Interactive and Gamified eBooks

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