FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS

- 10		Structure of functional	Example 6	of a compound
111	omologous series	group	Name	Structural formula
1000	alkanes	Only C-H and C-C single bonds	ethane .	н
	alkenes)c=c(ethene	>=<
	alkynes	-c≡c-	ethyne	н-с=с-н
	aloalkanes ilkyl halides)		bromoethane	H—C—C—Br
	alcohols (alkanols)	-¢-о-н	ethanol	H-C-C-0-K
	aldehydes	н_	ethanal	н—Ё—Ё—н
	ketones	_!_!_!_	proper-2-one	
ca	rboxylic acids	_0-0-н	ethanoic acid	н-с-с-о-н н
	esters	_8oţ-	methyl ethancate	H

NAMING OF ORGANIC COMPOUNDS

The name of every organic molecule has three parts:

- . The parent name indicates the number of C atoms in the longest carbon chain in the molecule.
- The suffix indicates what functional group is present.
- The prefix reveals the identity, location and number of substituents attached to the carbon. chain.

prefix. What and where are the substituents?

pairent How many carbons?

What is the functional group/homologous series?

sufflice

cla	-	·2.
$A_{L} = a$	-	3
156	ro.	- ep.

Number of carbon atoms	Parent name	Number of carbon atoms	Parent name
1	meth	5	pent
2	eth	6	hex
3	prop	7	hept
4	but	8	oct

prefix, suffix and parent!

Page 4 of 41 Tetra - chloro - melha

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