

Name: _____

Hour: _____

Temperature and Heat

21. _____ The flow of energy from an object at a higher temperature to an object at a lower temperature
22. _____ The total random kinetic energy of particles in an object
23. _____ The amount of energy needed to raise the temperature of one gram of water by 1°C
43. _____ The standard scientific unit in which energy is measured
53. _____ The metric temperature scale
63. _____ The SI unit for temperature
73. _____ The amount of energy required to raise the temperature of one gram of a substance by 1°C
83. _____ The process that moves energy from one object to another when they are touching physically
93. _____ Materials that transfer energy easily
103. _____ Are poor conductors that have a high specific heat
113. _____ All of the particles that make up matter are constantly in motion
123. _____ A measure of the average kinetic energy of all the particles in an object
133. _____ Temperature is measured in these units
143. _____ Measures temperature through the regular variation of some physical property of the material inside the thermometer
153. _____ The process that transfers energy by the movement of large numbers of particles in the same direction within a liquid or gas
163. _____ Energy that travels as the electromagnetic waves, it includes visible light, microwaves, and infrared light
173. _____ Change in volume of an object due to expansion caused by motion of particles as the temperature increases
183. _____ The standard temperature scale

Science Test Questions Heat And Temperature

RJ Shavelson



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