## WKS 7.1 - Beginning Naming & Formula Writing (2 pages)

#### Name the following ionic compounds. Be sure to check for Roman Numerals!

Na <sub>3</sub> P	MgBr <sub>2</sub>	
Ag <sub>2</sub> Se	AlCl <sub>3</sub>	
Ca <sub>3</sub> P <sub>2</sub>	Ba <sub>3</sub> N <sub>2</sub>	
Li <sub>2</sub> O	ZnF <sub>2</sub>	
K <sub>2</sub> S	SrI <sub>2</sub>	
(NH <sub>4</sub> ) <sub>2</sub> S	Ca <sub>3</sub> (PO <sub>3</sub> ) <sub>2</sub>	
Li <sub>2</sub> CO <sub>3</sub>	Al(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>3</sub>	
KCN	AgNO <sub>3</sub>	
Sb(NO <sub>3</sub> )s	NiS	
Cu(NO <sub>3</sub> ) <sub>2</sub>	FeCl <sub>2</sub>	
Au(CN)s	Pb(SO <sub>4</sub> ) <sub>2</sub>	

#### Write the correct formulas for each of the ionic compounds.

silver iodide	lithium sulfide	
magnesium nitride	cadmium bromide	
zinc chloride	barium phosphide	
aluminum fluoride	strontium oxide	
potassium permanganate	strontium sulfate	
cesium bromate	magnesium cyanide	
aluminum hypoiodite	cadmium periodate	
tin (IV) chlorate	bismuth (III) carbonate	
copper (I) sulfide	iron (III) oxide	

# **Pre Ap Chemistry Naming And Formula Writing Answer**

**Jin-Ying Zhang** 

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