

5 • Chemical Bonding**LEWIS STRUCTURES**Indicate the # of VALENCE electrons for each species. Write the correct Lewis electron dot structure for each.

F # of electrons = <u>7</u> $\cdot \ddot{\text{F}} \cdot$	O # of electrons = <u>6</u> $\cdot \ddot{\text{O}} \cdot$ or $\cdot \ddot{\text{O}} \cdot$	K # of electrons = <u>1</u> $\text{K} \cdot$	Al # of electrons = <u>3</u> $\cdot \text{Al} \cdot$
F⁻ <u>7+1</u> # of electrons = <u>8</u> $[\cdot \ddot{\text{F}} :]^{-}$	O²⁻ <u>6+2</u> # of electrons = <u>8</u> $[\cdot \ddot{\text{O}} :]^{2-}$	K⁺ <u>1-1</u> # of electrons = <u>0</u> $[\text{K}]^{+}$	Al³⁺ <u>3-3</u> # of electrons = <u>0</u> $[\text{Al}]^{3+}$
F₂ <u>7+7</u> # of electrons = <u>14</u> $\cdot \ddot{\text{F}} : \ddot{\text{F}} \cdot$	H₂ <u>1+1</u> # of electrons = <u>2</u> $\text{H} : \text{H}$	HF <u>1+7</u> # of electrons = <u>8</u> $\text{H} : \ddot{\text{F}} \cdot$	NH₃ <u>5+3</u> # of electrons = <u>8</u> $\begin{array}{c} \text{H} : \ddot{\text{N}} : \text{H} \\ \\ \text{H} \end{array}$
CH₄ <u>4+4</u> # of electrons = <u>8</u> $\begin{array}{c} \text{H} \\ \\ \text{H} : \text{C} : \text{H} \\ \\ \text{H} \end{array}$	NF₃ <u>5+21</u> # of electrons = <u>26</u> $\begin{array}{c} \cdot \ddot{\text{F}} : \ddot{\text{N}} : \ddot{\text{F}} : \\ \\ \cdot \ddot{\text{F}} : \end{array}$	SiF₄ <u>4+28</u> # of electrons = <u>32</u> $\begin{array}{c} \cdot \ddot{\text{F}} : \\ \cdot \ddot{\text{F}} : \text{Si} : \ddot{\text{F}} : \\ \\ \cdot \ddot{\text{F}} : \end{array}$	C₂H₆ <u>8+6</u> # of electrons = <u>14</u> $\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{H} : \text{C} : \text{C} : \text{H} \\ \quad \\ \text{H} \quad \text{H} \end{array}$
MgH₂ <u>2+2</u> # of electrons = <u>4</u> $\text{H} : \text{Mg} : \text{H}$	LiH <u>1+1</u> # of electrons = <u>2</u> $\text{Li} : \text{H}$	AlH₃ <u>3+3</u> # of electrons = <u>6</u> $\begin{array}{c} \text{H} : \text{Al} : \text{H} \\ \\ \text{H} \end{array}$	BH₃ <u>3+3</u> # of electrons = <u>6</u> $\begin{array}{c} \text{H} : \text{B} : \text{H} \\ \\ \text{H} \end{array}$

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Jicheng Xie



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Chemistry for the IB Diploma Exam Preparation Guide Steve Owen, Chris Martin, 2015-06-25 Chemistry for the IB Diploma Second edition covers in full the requirements of the IB syllabus for Chemistry for first examination in 2016

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problems are incorporated into the body of the text in an effort to engage students more directly in chemistry Arrowless mechanisms seek to help students develop their electron pushing skills and intuition through repeated practice By design this volume is more actively engaging than a traditional textbook In addition the historical development of ideas is presented to help frame and center these concepts for the reader Primary and summative sources are given for all topics covered The sources provide definitive information for the reader and ensure that all information is supported by peer reviewed experimental sources Features The development of key ideas is presented in their historical context All information presented is supported through citations to chemical literature Problems are incorporated into the body of the text including arrowless mechanisms which encourage students to engage more actively and to develop their electron pushing skills and intuition International Union of Pure and Applied Chemistry style and technical guidelines are followed throughout the text The problems text and presentation are based on years of classroom refinement of teaching pedagogy **Chemical Bonding** Mark J. Winter, 2016 The renowned Oxford Chemistry Primers series which provides focused introductions to a range of important topics in chemistry has been refreshed and updated to suit the needs of today's students lecturers and

postgraduate researchers The rigorous yet accessible treatment of each subject area is ideal for those wanting a primer in a given topic to prepare them for more advanced study or research The learning features provided including questions at the end of every chapter and online multiple choice questions encourage active learning and promote understanding Furthermore frequent diagrams margin notes and glossary definitions all help to enhance a student s understanding of these essential areas of chemistry Chemical bonding gives a clear and succinct explanation of this fundamental topic which underlies the structure and reactivity of all molecules and therefore the subject of chemistry itself Little prior knowledge or mathematical ability is assumed making this the perfect text to introduce students to the subject **NEET UG Physics**

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later at diego rivera's side they hosted a fascinating array of luminaries from Mexico and abroad drawn by the charisma of both artists. Frida and Diego wanted to leave her house as a museum for all Mexicans to enjoy.

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web kahlo stated that she was born at the family home la casa azul the blue house but according to the official birth registry the birth took place at the nearby home of her maternal grandmother kahlo s parents were photographer guillermo kahlo 1871 1941 and matilde calderón y gonzález 1876 1932 and they were thirty six and thirty

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there she was born lived worked fell in love forgave and even hated diego rivera today this typical coyoacan mansion is also one of the most visited museums in mexico city along with the national museum of anthropology and

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