

Periodic Trends

1. Explain the following trends:

- Atomic Radius
- Electronegativity
- Ionization Energy

2. Order the following elements by increasing atomic radius:

Li, Na, K, Rb, Cs, Fr

3. Order the following elements by increasing electronegativity:

F, Cl, Br, I, At

4. Order the following elements by increasing ionization energy:

Li, Na, K, Rb, Cs, Fr

5. Order the following elements by increasing metallic character:

Li, Na, K, Rb, Cs, Fr

6. Order the following elements by increasing non-metallic character:

F, Cl, Br, I, At

7. Order the following elements by increasing atomic mass:

Li, Na, K, Rb, Cs, Fr

8. Order the following elements by increasing atomic number:

Li, Na, K, Rb, Cs, Fr

9. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

10. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

11. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

12. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

13. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

14. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

15. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

Name: _____ Date: _____ Period: _____

Periodic Trends Test Study Guide

Section 1: Atomic Structure and Properties of the Periodic Table

1. What are the three main categories of elements in the periodic table?

- Metals
- Non-metals
- Metalloids

2. List the following elements in order of increasing atomic radius: Li, Na, K, Rb, Cs, Fr.

3. List the following elements in order of increasing electronegativity: F, Cl, Br, I, At.

4. List the following elements in order of increasing ionization energy: Li, Na, K, Rb, Cs, Fr.

5. List the following elements in order of increasing metallic character: Li, Na, K, Rb, Cs, Fr.

6. List the following elements in order of increasing non-metallic character: F, Cl, Br, I, At.

7. List the following elements in order of increasing atomic mass: Li, Na, K, Rb, Cs, Fr.

8. List the following elements in order of increasing atomic number: Li, Na, K, Rb, Cs, Fr.

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15. List the following elements in order of increasing atomic weight: Li, Na, K, Rb, Cs, Fr.

Science Study Guide Periodic Table

Christian Drosten



Science Study Guide Periodic Table:

Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Science Study Guide Periodic Table**

In a world inundated with screens and the cacophony of quick connection, the profound energy and mental resonance of verbal artistry often disappear into obscurity, eclipsed by the regular barrage of noise and distractions. However, located within the musical pages of **Science Study Guide Periodic Table**, a charming work of fictional splendor that impulses with fresh feelings, lies an unforgettable trip waiting to be embarked upon. Penned by a virtuoso wordsmith, this magical opus guides viewers on a mental odyssey, gently exposing the latent potential and profound influence stuck within the complex web of language. Within the heart-wrenching expanse of the evocative examination, we shall embark upon an introspective exploration of the book's main themes, dissect its captivating writing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers' souls.

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