Anatomy and Physiology (Sample Questions- Chapter 2)

Name of the Student:

Question 1. Sodium has an atomic number of 11 and an atomic mass of 23. Sodium has _______.

A. 12 neutrons and 11 protons

- B. 12 protons and 11 neutrons
- C. 12 electrons and 11 neutrons
- D. 12 protons and 11 electrons
- E. 12 electrons and 11 protons

Question 2. What type of bond attracts one water molecule to another?

- A. An ionic bond
- B. A peptide bond

C. A hydrogen bond

- D. A covalent bond
- E. A hydrolytic bond

Question 3. Which of these has the highest H' concentration?

A. Lemon juice, pH = 2.3

- B. Red wine, pH = 3.2
- C. Tomato juice, pH = 4.7
- D. Saliva, pH = 6.6
- E. Household ammonia, pH = 10.8

Ques 4. An atom is made up of protons, neutrons, and electrons. The proton carries the following charge:

- A. negative
- B. no charge
- C. positive
- D. none of the above

Question 5. In the following reaction, what is(are) the product(s)? CO2 + H2O ---> H2CO3

A. HyCO:

- B. CO₂ and H₂O
- C. CO₂ and H₂CO₃.
- D. H₂O and H₂CO₃

Sample Questions Chapter With Answers

Camilla Rothe

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