Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H₂ hydrogen is distornic

2.) F2 discripe in discrepie

$$F + F \rightarrow F \rightarrow F$$

3.) O2 oxygen is discomic

$$O + O \rightarrow$$

$$\longrightarrow$$
 O=O

Double Bond

4.) No nitrogen is diatomic, Is this a triple bond?

$$N + N$$

5.) BF3 you need 3 fluoring atoms here

$$B + F$$

6.) Ammonia NH3 hint how many hydrogen atoms are needed?

$$N + H$$

7.) Carbon dioxide CO-

$$O + C + O$$

8.) Methane CH4 careful here 4 hydrogen atoms needed

$$C + H$$

9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!

$$H + O + H$$

10.) SO₂ hint: one pair of electrons from sulfar must be slit up for this one to work.

$$O + S + O$$

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LP Steffe

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