

Name _____ **answers** _____ Date _____

FORMULAS WITH POLYTOMIC IONS

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions.

| | OH^- | OH_3^- | CO_3^{2-} | SO_4^{2-} | PO_4^{3-} |
|------------------|--|----------------------------|------------------------------|------------------------------|------------------------------|
| H^+ | HOH (H_2O) | HNO_3 | H_2CO_3 | H_2SO_4 | H_3PO_4 |
| Na^+ | NaOH | NaNO_3 | Na_2CO_3 | Na_2SO_4 | Na_3PO_4 |
| Mg^{2+} | $\text{Mg}(\text{OH})_2$ | $\text{Mg}(\text{NO}_3)_2$ | MgCO_3 | MgSO_4 | $\text{Mg}_3(\text{PO}_4)_2$ |
| NH_4^+ | NH_4OH | NH_4NO_3 | $(\text{NH}_4)_2\text{CO}_3$ | $(\text{NH}_4)_2\text{SO}_4$ | $(\text{NH}_4)_3\text{PO}_4$ |
| Ca^{2+} | KOH | KNO_3 | K_2CO_3 | K_2SO_4 | K_3PO_4 |
| K^+ | $\text{Al}(\text{OH})_3$ | $\text{Al}(\text{NO}_3)_3$ | $\text{Al}_2(\text{CO}_3)_3$ | $\text{Al}_2(\text{SO}_4)_3$ | AlPO_4 |
| Al^{3+} | $\text{Pb}(\text{OH})_4$ | $\text{Pb}(\text{NO}_3)_4$ | $\text{Pb}(\text{CO}_3)_2$ | $\text{Pb}(\text{SO}_4)_2$ | $\text{Pb}_3(\text{PO}_4)_4$ |

Polyatomic Ionic Formulas Answers

G Orfield



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