

Name: Michaela Huber

Date:

Class: Physical Sciences

Physical Sciences Chapter 10: Heat, Thermal Energy and Work

Multiple Choice

1. The first law of thermodynamics is another way of stating:
a) conservation of energy
b) conservation of mass
c) conservation of momentum
d) conservation of charge
2. Energy added to a system
a) results in a change in the system's internal energy
b) is lost to the surroundings
c) is converted to work
d) is converted to kinetic energy of the system
3. When a definite quantity of heat is added to an object,
a) the object's mass increases
b) the energy is lost to the system
c) the object is 100% efficient
d) it gains energy from the surroundings
4. A heat engine transforms thermal energy mainly by
a) conduction
b) convection
c) radiation
d) expansion and contraction
5. Compression of a gas results in
a) cooling
b) heating
c) evaporation
d) condensation
6. Why does liquid rise in a thermometer?
a) conduction
b) convection
c) expansion
d) radiation
7. Which statement is NOT true about heat?
a) It is a scalar quantity.
b) Heat always flows from higher to lower temperature.
c) Heat always flows from lower to higher temperature.
d) It is a form of energy transfer between systems.
8. Which of the following is a thermal conductor?
a) aluminum
b) wood
c) air
d) foam

Physical Science Chapter 16 Section 1 Answers

Clifford Lane



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