

FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS

Homologous series	Structure of functional group	Example of a compound	
		Name	Structural formula
alkanes	Only C-H and C-C single bonds	ethane	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{H} \\ & & \\ \text{H} & & \text{H} \end{array}$
alkenes	$\begin{array}{c} \diagup & & \diagdown \\ & \text{C}=\text{C} & \\ \diagdown & & \diagup \end{array}$	ethene	$\begin{array}{c} \text{H} & & \text{H} \\ & \diagdown & \diagup \\ & \text{C}=\text{C} & \\ & \diagup & \diagdown \\ \text{H} & & \text{H} \end{array}$
alkynes	$-\text{C}\equiv\text{C}-$	ethyne	$\text{H}-\text{C}\equiv\text{C}-\text{H}$
haloalkanes (alkyl halides)	$\begin{array}{c} \\ -\text{C}-\text{X} \\ \\ \text{(X = F, Cl, Br, I)} \end{array}$	bromoethane	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{Br} \\ & & \\ \text{H} & & \text{H} \end{array}$
alcohols (alkanols)	$\begin{array}{c} \\ -\text{C}-\text{O}-\text{H} \\ \end{array}$	ethanol	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}-\text{H} \\ & & \\ \text{H} & & \text{H} \end{array}$
aldehydes	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{H} \end{array}$	ethanal	$\begin{array}{c} \text{H} & & \text{O} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{H} \\ \\ \text{H} \end{array}$
ketones	$\begin{array}{c} \text{O} \\ \\ -\text{C}- \\ \end{array}$	propan-2-one	$\begin{array}{c} \text{H} & & \text{O} & & \text{H} \\ & & & & \\ \text{H}-\text{C} & - & \text{C} & - & \text{C}-\text{H} \\ & & & & \\ \text{H} & & & & \text{H} \end{array}$
carboxylic acids	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{O}-\text{H} \end{array}$	ethanoic acid	$\begin{array}{c} \text{H} & & \text{O} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}-\text{H} \\ \\ \text{H} \end{array}$
esters	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{O}-\text{C}- \\ \end{array}$	methyl ethanoate	$\begin{array}{c} \text{H} & & \text{O} & & \text{H} \\ & & & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}- & \text{C}-\text{H} \\ & & & & \\ \text{H} & & & & \text{H} \end{array}$

NAMING OF ORGANIC COMPOUNDS

The name of every organic molecule has three parts:

- The **parent name** indicates the number of C atoms in the longest carbon chain in the molecule.
- The **suffix** indicates what functional group is present.
- The **prefix** reveals the identity, location and number of substituents attached to the carbon chain.

prefix

What and where are the substituents?

parent

How many carbons?

suffix

What is the functional group/homologous series?

di - 2

tri - 3

tetra - 4

Number of carbon atoms	Parent name	Number of carbon atoms	Parent name
1	meth	5	pent
2	eth	6	hex
3	prop	7	hept
4	but	8	oct



prefix, suffix and parent!

Page 4 of 41
Tetra-chloro-metha

Physical Science Organic Compounds Study Guide

**James Adams, Maryn Munley, Nathan
Eisenbeis, Caroline Miller, Jesse Shetler**



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Table of Contents Physical Science Organic Compounds Study Guide

1. Understanding the eBook Physical Science Organic Compounds Study Guide
 - The Rise of Digital Reading Physical Science Organic Compounds Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Organic Compounds Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Organic Compounds Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Organic Compounds Study Guide

- Personalized Recommendations
- Physical Science Organic Compounds Study Guide User Reviews and Ratings
- Physical Science Organic Compounds Study Guide and Bestseller Lists
- 5. Accessing Physical Science Organic Compounds Study Guide Free and Paid eBooks
 - Physical Science Organic Compounds Study Guide Public Domain eBooks
 - Physical Science Organic Compounds Study Guide eBook Subscription Services
 - Physical Science Organic Compounds Study Guide Budget-Friendly Options
- 6. Navigating Physical Science Organic Compounds Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Organic Compounds Study Guide Compatibility with Devices
 - Physical Science Organic Compounds Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Organic Compounds Study Guide
 - Highlighting and Note-Taking Physical Science Organic Compounds Study Guide
 - Interactive Elements Physical Science Organic Compounds Study Guide
- 8. Staying Engaged with Physical Science Organic Compounds Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Organic Compounds Study Guide
- 9. Balancing eBooks and Physical Books Physical Science Organic Compounds Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Organic Compounds Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Organic Compounds Study Guide
 - Setting Reading Goals Physical Science Organic Compounds Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Organic Compounds Study Guide

- Fact-Checking eBook Content of Physical Science Organic Compounds Study Guide
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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