

Reviewing Content

42. The solvent is the substance in which the solute is dissolved.
43. Random collisions of the solvent molecules with the solute particles provide enough force to overcome gravity.
44. **solubility**: the amount of a substance that dissolves in a given quantity of solvent at specified conditions of temperature and pressure to produce a saturated solution.
saturated solution: a solution containing the maximum amount of solute for a given amount of solvent at a constant temperature and pressure. **unsaturated solution**: a solution that contains less solute than a saturated solution at a given temperature and pressure. **miscible** describes liquids that dissolve in each other. **immiscible** describes liquids that are insoluble in each other.
45. Particles of solute crystallize.
46. No; if there were undissolved solute, the excess solute would come out of a supersaturated solution.
47. 5.55×10^2 g AgNO_3
48. Solubility increases with pressure.
49. a. 1.6×10^{-2} g/L
b. 4.7×10^{-2} g/L
50. Dilute and concentrated are relative terms and are not quantitative. Molarity provides the exact number of moles of solute per liter of solution.
51. Molarity is the number of moles of solute dissolved in one liter of solution.
a. 1.3M KCl
b. 5.3×10^{-2} M MgCl_2
52. 2.00×10^2 mL
53. a. 5.0×10^{-2} mol NaCl, 20 g NaCl
b. 1.0 mol KNO_3 , 1.0×10^2 g KNO_3
c. 2.5×10^2 mol CaCl_2 , 2.0 g CaCl_2
54. a. 2.3×10^2 g NaCl
b. 2.0 g MgCl_2
55. a. 10% (v/v) ethanol
b. 63.6% (v/v) isopropyl alcohol
56. Colligative properties are properties of a solution that depend only on the number of solute particles; boiling-point elevation, freezing-point depression, and vapor-pressure lowering. Boiling points are elevated because shells of solvent form around solute particles, reducing the amount of solvent molecules that have sufficient energy to escape the solution; relative to the pure solvent, the amount of energy required to cause vaporization or boiling increases. Solutes disrupt the ordering of the solvent structure, so more kinetic energy must be withdrawn from a solution for it to solidify. This lowers the freezing point of the solution.
57. a. sea water
b. 1.5M KNO_3
c. 0.100M MgCl_2
58. The effective molality of the $\text{Ca}(\text{NO}_3)_2$ solution is 3m. The effective molality of the NaNO_3 solution is 2m.
59. When vapor pressure is lowered relative to pure solvent, more energy must be supplied to reach the boiling point; thus the boiling point is increased relative to pure solvent.
60. The salt lowers the freezing point of the ice-water cooling mixture.
61. 1M solution: 1 mol of solute in 1 L of solution; 1m solution: 1 mol of solute in 1000 g of solvent
62. Add 27.0 g H_2O to 32.0 g $\text{C}_2\text{H}_5\text{OH}$.
63. a. -100.26°C
b. -101.54°C
64. a. -4.48°C
b. -2.2°C
65. a. -1.1°C
b. -0.24°C
c. -1.3°C

Understanding Concepts

66. a. The freezing-point depression is twice as great for solute B; solute B must provide twice as many particles in solution.

Prentice Hall Chemistry 16 Practice Problem Answers

Jessica J Manson



Prentice Hall Chemistry 16 Practice Problem Answers:

Catalog of Copyright Entries. Third Series Library of Congress. Copyright Office, 1956 Includes Part 1 Number 2 Books and Pamphlets Including Serials and Contributions to Periodicals July December *Learning with Understanding in the Chemistry Classroom* Iztok Devetak, Saša Aleksij Glažar, 2014-01-14 This volume offers a critical examination of a variety of conceptual approaches to teaching and learning chemistry in the school classroom Presenting up to date research and theory and featuring contributions by respected academics on several continents it explores ways of making knowledge meaningful and relevant to students as well as strategies for effectively communicating the core concepts essential for developing a robust understanding of the subject Structured in three sections the contents deal first with teaching and learning chemistry discussing general issues and pedagogical strategies using macro sub micro and symbolic representations of chemical concepts Researchers also describe new and productive teaching strategies The second section examines specific approaches that foster learning with understanding focusing on techniques such as cooperative learning presentations laboratory activities multimedia simulations and role playing in forensic chemistry classes The final part of the book details learner centered active chemistry learning methods active computer aided learning and trainee chemistry teachers use of student centered learning during their pre service education Comprehensive and highly relevant this new publication makes a significant contribution to the continuing task of making chemistry classes engaging and effective Catalogue of Title-entries of Books and Other Articles Entered in the Office of the Librarian of Congress, at Washington, Under the Copyright Law ... Wherein the Copyright Has Been Completed by the Deposit of Two Copies in the Office Library of Congress. Copyright Office, 1978 **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office, 1966 **Books in Print** ,1954 *Nanoparticulate Drug Delivery Systems* Deepak Thassu, Michel Deleers, Yashwant Vishnupant Pathak, 2007-03-30 With the advent of analytical techniques and capabilities to measure particle sizes in nanometer ranges there has been tremendous interest in the use of nanoparticles for more efficient methods of drug delivery Nanoparticulate Drug Delivery Systems addresses the scientific methodologies formulation processing applications recent trends and e **The Journal of Education** ,1893 *Catalogue of the School Bulletin [and New York State Educational Journal]* Publications ,1908 *Catalogue of the School Bulletin Publications* Charles William Bardeen, 1903 Journal of Education ,1882

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our mission is focused on forging social development through the arts which we carry out through five key programs residencies exhibitions clay films and mobile library casa wabi is located on the pacific coast 30 minutes from the puerto escondido airport oaxaca

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