FLAME TEST LAB PROCEDURE

Materials & Equipment

Safety Goggles Gloves Beakers, 250-mL, 2 Matches Weighing dishes, 6 Wooden splints soaked in water, 6 Water, distilled or deionized, 250 mL

Calcium chloride, CaCl2_2H2O, 1–1.5 g Copper chloride, CuCl2_2H2O, 1–1.5 g Lithium chloride, LiCl, 1–1.5 g Potassium chloride, KCl, 1–1.5 g Sodium chloride, NaCl, 1–1.5 g Strontium chloride, SrCl2_6H2O, 1–1.5 g

Procedure:

- 1. Put on Safety Goggles.
- 2. Wear gloves when handling chemicals.
- Obtain a 250-mi, beaker about half-full with distilled or deionized water. Obtain six wooden splints that have been soaked in distilled or deionized water. Place them in this beaker of water to continue soaking at your lab station.
- 4. Fill a second 250-mL beaker about half-full with tap water. Label this beaker "rinse water".
- Obtain six weighing dishes labeled "CaCl2", "CuCl2", "LiCl", "NaCl", "KCl", "SrCl2". Each contains a small scoopful (about 1–1.5 g) of each metallic solid in the corresponding weighing dish.
- 6. Dip the soaked end of one of the wooden splints in one of the metallic salts.
- Light a match and then place the "dipped" wooden splint into the flame. Observe the color of the flame. Allow the splint to burn until the color fades. Try not to allow any of the solid to fall onto your desk. If necessary, repeat the test with the same splint and salt.
- Immerse the wooden splint in the "rinse water" to fully extinguish it, as well as the match. Then
 discard both into the trash.
- 9. Record your observations for the flame color produced by the metallic salt in the Data Table.
- Repeat Steps 6–8 for the other five metallic salts. Record your observations for the flame color produced by each metallic salt in the Data Table.
- 11. Clean up your area, put everything away, and wash your hands before leaving the lab.

Representative Wavelength, nm	Wavelength Region, nm	Color
410	400-425	Violet
470	425-480	Blue
490	480-500	Blue-green
520	500-560	Green
565	560-580	Yellow-green
580	580-585	Yellow
600	585-650	Orange
650	650-700	Red

Figure 1.

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TD Snyder

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Table of Contents Prentice Hall Flame Test Lab 10 Answers

- 1. Understanding the eBook Prentice Hall Flame Test Lab 10 Answers
 - The Rise of Digital Reading Prentice Hall Flame Test Lab 10 Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Prentice Hall Flame Test Lab 10 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Flame Test Lab 10 Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Prentice Hall Flame Test Lab 10 Answers
 - Personalized Recommendations
 - Prentice Hall Flame Test Lab 10 Answers User Reviews and Ratings
 - Prentice Hall Flame Test Lab 10 Answers and Bestseller Lists
- 5. Accessing Prentice Hall Flame Test Lab 10 Answers Free and Paid eBooks
 - Prentice Hall Flame Test Lab 10 Answers Public Domain eBooks
 - Prentice Hall Flame Test Lab 10 Answers eBook Subscription Services
 - Prentice Hall Flame Test Lab 10 Answers Budget-Friendly Options
- 6. Navigating Prentice Hall Flame Test Lab 10 Answers eBook Formats

- o ePub, PDF, MOBI, and More
- Prentice Hall Flame Test Lab 10 Answers Compatibility with Devices
- Prentice Hall Flame Test Lab 10 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Prentice Hall Flame Test Lab 10 Answers
 - Highlighting and Note-Taking Prentice Hall Flame Test Lab 10 Answers
 - Interactive Elements Prentice Hall Flame Test Lab 10 Answers
- 8. Staying Engaged with Prentice Hall Flame Test Lab 10 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Flame Test Lab 10 Answers
- 9. Balancing eBooks and Physical Books Prentice Hall Flame Test Lab 10 Answers
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Prentice Hall Flame Test Lab 10 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Prentice Hall Flame Test Lab 10 Answers
 - Setting Reading Goals Prentice Hall Flame Test Lab 10 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Prentice Hall Flame Test Lab 10 Answers
 - Fact-Checking eBook Content of Prentice Hall Flame Test Lab 10 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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