

Physical Chemistry 2nd Edition Ball

Solutions Manual

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Chapter 7

Equilibria for Multiple Component Systems

- 7.2. With only a single component and phase, if only $\text{Fe}_2(\text{SO}_4)_3$ were present in our system, we would have $C - P + 2 = 1 - 1 + 2 = 2$ degrees of freedom.
- 7.4. For a single-component system, there will never be negative degrees of freedom. The maximum number of phases that can exist in equilibrium is three, so the Gibbs phase rule would yield $1 - 3 + 2 = 0$ degrees of freedom (like might occur at the triple point). For a two-component system, the maximum number of phases that can exist in equilibrium is four. There will never be negative degrees of freedom.
- 7.6. In this reaction, there are only three independent components (the amount of the fourth can be determined from the equilibrium constant) and 3 phases, two solids and a gas mixture. Therefore, there are $3 - 3 + 2 = 2$ degrees of freedom.
- 7.8. For water: $n = \frac{pV}{RT} = \frac{(23.76 \text{ torr})(5.00 \text{ L})}{(0.08205 \text{ L} \cdot \text{atm/mol} \cdot \text{K})(298.15 \text{ K})} \times \frac{1 \text{ atm}}{760 \text{ torr}} = 6.39 \times 10^{-3} \text{ mol}$ of water is needed to ensure that there is a liquid and gas phase. This equals 0.115 grams.
For methanol:
- $$n = \frac{pV}{RT} = \frac{(125.0 \text{ torr})(5.00 \text{ L})}{(0.08205 \text{ L} \cdot \text{atm/mol} \cdot \text{K})(298.15 \text{ K})} \times \frac{1 \text{ atm}}{760 \text{ torr}} = 3.36 \times 10^{-2} \text{ mol}$$
- of methanol is needed to ensure that there is a liquid and gas phase. This equals 1.08 grams.
- 7.10. Since, according to equation 7.11, $a_i = \frac{p_i}{p_i^*}$, and since $p_i^* = 760 \text{ torr}$, we can calculate a_i :

Physical Chemistry Ball Manual

J Elliott



Physical Chemistry Ball Manual:

Whispering the Secrets of Language: An Emotional Journey through **Physical Chemistry Ball Manual**

In a digitally-driven world wherever screens reign great and quick communication drowns out the subtleties of language, the profound secrets and psychological nuances hidden within words often move unheard. However, nestled within the pages of **Physical Chemistry Ball Manual** a charming fictional prize pulsing with fresh emotions, lies a fantastic quest waiting to be undertaken. Written by a skilled wordsmith, that charming opus invites readers on an introspective journey, lightly unraveling the veiled truths and profound influence resonating within the fabric of each word. Within the mental depths with this poignant review, we can embark upon a heartfelt exploration of the book's primary themes, dissect their fascinating publishing fashion, and fail to the strong resonance it evokes serious within the recesses of readers' hearts.

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