Name:	KEY	
Hourt	Date:	

Chemistry - Review Chapters (415) 6

Directions: Answer each of the following questions. You need not use complete sentences.

By what properly did Mendeleev arrange the elements?

ATOMIC MASS

2. By what property did Moseley suggest that the periodic table be arranged?

ATOMIC NUMBER

3. What is a period? How many are there in the periodic table?

A HORIZONTAL ROW IN THE PERIODIC TABLE: 7

4. What is a group (also called a family)? How many are there in the periodic table?

A VERTICAL COLUMN IN THE PERIODIC TABLE: 18

State the number of valence electrons in an atom of:

a. sulfur	b. calcium	c. chilorine	d. arsenic
6	2	7	5

Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10	b. 18	c. 36	d. 90
Ne. NEON	Ar, ARGON	Kr, KRYPTON	Th; THORNUM

List, by number, both the period and group of each of these elements.

	Symbol	Period	Group
a. beryllium	Be	2	2
b. iron	Fo	4	8
c. lead	Pb	6	14

8. Which of the following pairs of elements belong to the same period?

a. Na and CI	b. Ne and Li	c. Na and Cu	d. Na and Ne
	THE RESERVE THE PARTY NAMED IN	THE RESERVE THE PERSON NAMED IN	TOTAL PLANTAGE SERVICES IN THESE

9. Which of the following pairs of elements belong to the same group?

a. H and He b. Li and Se c. C and Pb d. Ga and Ge

10. How does an element's period number relate to the number of the energy level of its valence electrons?

PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

11. What are the transition elements?

GROUPS 3-12

12. In what type of orbitals are the actinide and lanthanide electrons found?

FORBITALS.

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Annelies Wilder-Smith

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