

11-3 Practice Problems

1. Identify the limiting reactant when 1.22-g of O_2 reacts with 1.65-g of H_2 to produce water.

2. Identify the limiting reactant when 4.68-g of Fe reacts with 2.88-g of S to produce FeS .

3. Identify the limiting reactant when 2.87-g of $\text{Mg}(\text{OH})_2$ reacts with 12.84-g of HCl to form MgCl_2 and water.

4. Identify the limiting reactant when 6.25-g of AgNO_3 reacts with 4.12-g of NaCl to form NaNO_3 and AgCl .

5. Identify the limiting reactant when 7.81-g of HCl reacts with 5.24-g of NaOH to produce NaCl and H_2O .

6. Identify the limiting reactant when 6.33-g of H_2SO_4 reacts with 5.92-g of NaOH to produce Na_2SO_4 and water.

7. Identify the limiting reactant when 49.25-g of CaC_2 reacts with 33.71-g of water to produce $\text{Ca}(\text{OH})_2$ and C_2H_2 .

8. Identify the limiting reactant when 63.14-g of CaCl_2 reacts with 74.68-g of Na_2CO_3 to produce CaCO_3 and NaCl .

9. Identify the limiting reactant when 4.687-g of SF_4 reacts with 6.281-g of I_2O_5 to produce IF_5 and SO_2 .

10. If 4.1-g of Cr is heated with 9.3-g of Cl_2 , what mass CrCl_3 will be produced?

11. What mass of SO_2 is produced from the reaction between 31.5-g of S_8 and 8.65-g of O_2 ?

12. What mass of SO_2 is produced from the reaction of 12.4-g of SO_2 and 3.45-g of O_2 ?

13. What mass of H_2SO_4 is produced from the reaction of 6.58-g of SO_3 and 1.64-g of H_2O ?

14. What mass of CdS is produced if 8.47-g of cadmium reacts with 2.51-g of Sulfur?

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Table of Contents Prentice Hall Chemistry Practice Problem

1. Understanding the eBook Prentice Hall Chemistry Practice Problem
 - The Rise of Digital Reading Prentice Hall Chemistry Practice Problem
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry Practice Problem
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Chemistry Practice Problem
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry Practice Problem
 - Personalized Recommendations
 - Prentice Hall Chemistry Practice Problem User Reviews and Ratings
 - Prentice Hall Chemistry Practice Problem and Bestseller Lists
5. Accessing Prentice Hall Chemistry Practice Problem Free and Paid eBooks
 - Prentice Hall Chemistry Practice Problem Public Domain eBooks
 - Prentice Hall Chemistry Practice Problem eBook Subscription Services
 - Prentice Hall Chemistry Practice Problem Budget-Friendly Options
6. Navigating Prentice Hall Chemistry Practice Problem eBook Formats

- ePub, PDF, MOBI, and More
- Prentice Hall Chemistry Practice Problem Compatibility with Devices
- Prentice Hall Chemistry Practice Problem Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Prentice Hall Chemistry Practice Problem
 - Highlighting and Note-Taking Prentice Hall Chemistry Practice Problem
 - Interactive Elements Prentice Hall Chemistry Practice Problem
- 8. Staying Engaged with Prentice Hall Chemistry Practice Problem
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Chemistry Practice Problem
- 9. Balancing eBooks and Physical Books Prentice Hall Chemistry Practice Problem
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Chemistry Practice Problem
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Prentice Hall Chemistry Practice Problem
 - Setting Reading Goals Prentice Hall Chemistry Practice Problem
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Prentice Hall Chemistry Practice Problem
 - Fact-Checking eBook Content of Prentice Hall Chemistry Practice Problem
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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