

Study Guide

Key Concepts

14.1 Properties of Gases

- Gases are easily compressed because of the space between particles in a gas.
- The amount of gas (n), volume (V), and temperature (T) are factors that affect gas pressure (P).

14.2 The Gas Laws

- As the pressure of a gas increases, the volume decreases, if the temperature is constant.
- As the temperature of an enclosed gas increases, the volume increases, if the pressure is constant.
- As the temperature of an enclosed gas increases, the pressure increases, if the volume is constant.

- The combined gas law allows you to do calculations for situations in which only the amount of gas is constant.

14.3 Ideal Gases

- To calculate the number of moles of a contained gas requires an expression that contains the variable n .
- Real gases differ most from an ideal gas at low temperatures and high pressures.

14.4 Gases: Mixtures and Movements

- In a mixture of gases, the total pressure is the sum of the partial pressures of the gases.
- Gases of lower molar mass diffuse and effuse faster than gases of higher molar mass.

Vocabulary

- Boyle's law (p. 418)
- Charles's law (p. 420)
- combined gas law (p. 424)
- compressibility (p. 413)
- Dalton's law of partial pressures (p. 432)
- diffusion (p. 435)
- effusion (p. 435)
- Gay-Lussac's law (p. 422)
- Graham's law of effusion (p. 435)
- ideal gas constant (p. 426)
- ideal gas law (p. 426)
- partial pressure (p. 432)

Key Equations

- Boyle's law: $P_1 \times V_1 = P_2 \times V_2$
- Charles's law: $\frac{V_1}{T_1} = \frac{V_2}{T_2}$
- Gay-Lussac's law: $\frac{P_1}{T_1} = \frac{P_2}{T_2}$
- Combined gas law: $\frac{P_1 \times V_1}{T_1} = \frac{P_2 \times V_2}{T_2}$
- Ideal gas law: $P \times V = n \times R \times T$ or $PV = nRT$
- Dalton's law: $P_{\text{total}} = P_1 + P_2 + P_3 + \dots$
- Graham's law: $\frac{\text{Rate}_A}{\text{Rate}_B} = \sqrt{\frac{\text{molar mass}_B}{\text{molar mass}_A}}$

Organizing Information

Use these terms to construct a concept map that organizes the major ideas of this chapter.



Concept Map 14 Solve the Concept Map with the help of an interactive guided tutorial.

with **ChemASAP**

Boyle's law

Charles's law

partial pressure

combined gas law

diffusion

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S Marginson



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who want to learn more about their field without unnecessary complexity detail or formalism *Digital Simulation in Electrochemistry* Dieter Britz, Jörg Strutwolf, 2016-05-09 This book explains how the partial differential equations pdes in electroanalytical chemistry can be solved numerically It guides the reader through the topic in a very didactic way by first introducing and discussing the basic equations along with some model systems as test cases systematically Then it outlines basic numerical approximations for derivatives and techniques for the numerical solution of ordinary differential equations Finally more complicated methods for approaching the pdes are derived The authors describe major implicit methods in detail and show how to handle homogeneous chemical reactions even including coupled and nonlinear cases On this basis more advanced techniques are briefly sketched and some of the commercially available programs are discussed In this way the reader is systematically guided and can learn the tools for approaching his own electrochemical simulation problems This new fourth edition has been carefully revised updated and extended compared to the previous edition Lecture Notes in Physics Vol 666 It contains new material describing migration effects as well as arrays of ultramicroelectrodes It is thus the most comprehensive and didactic introduction to the topic of electrochemical simulation *The Handbook of Board Governance* Richard Leblanc, 2020-04-06 The revised new edition of the must read guide for executives provides comprehensive coverage of topics in corporate governance by leading subject matter experts The Handbook of Board Governance is the marketing leading text on public nonprofit and private board governance Providing comprehensive in depth coverage this unique text represents a collaboration of internationally recognized academics and prominent organization directors executives managers and advisors Contributors include Ariel Fromer Babcock Robert Eccles Alice Korngold Ellie Mulholland Michael Useem Elizabeth Valentine and John Zinkin Practical expert guidance enables readers to understand value creation and the strategic role of the board risk governance and oversight audit and compensation committee effectiveness CEO succession planning and other diverse board duties and responsibilities Now in its second edition the Handbook offers substantial updates and revisions reflecting contemporary trends practices and developments in board governance New content includes discussions of pressing issues related to climate change examination of information technology and cybersecurity challenges and recent tax legislation that will impact executive compensation Editor Dr Richard Leblanc an award winning teacher professor lawyer management consultant and specialist on boards of directors integrates practical experience and academic rigor to assist readers Build and strengthen engaged and collaborative leadership in the boardroom Recognize the role and responsibilities of a well functioning governing board Risk governance assurance and the duties of directors Keep pace with new trends in board governance and shareholder responsibility Measure performance and align performance measurement to executive pay Understand information technology governance sustainability governance and the different forms of governance Highly relevant to board and committee members regardless of sector or industry The Handbook of Board Governance 2nd Edition is an invaluable source of knowledge on all aspects of

corporate and organization governance *Study Guide and Solutions Manual for Organic Chemistry* Susan McMurry, John McMurry, 1988 **Chemistry and Physics of Aqueous Gas Solutions** William Alfred Adams, 1975 **Modern Electrochemistry** John O'M. Bockris, Amulya K.N. Reddy, 1973-10-31

7 The Electrified Interface 7 1 Electrification of an Interface 7 1 1 The Electrode Electrolyte Interface The Basis of Electrodics 7 1 2 New Forces at the Boundary of an Electrolyte 7 1 3 The Interphase Region Has New Properties and New Structures 7 1 4 An Electrode Is Like a Giant Central Ion 7 1 5 The Consequences of Compromise Arrangements The Electrolyte Side of the Boundary Acquires a Charge 7 1 6 Both Sides of the Interface Become Electrified The So Called Electrical Double Layer 7 1 7 Double Layers Are Characteristic of All Phase Boundaries 7 1 8 A Look into an El

Characterization of Cavitation Bubbles and Sonoluminescence Rachel Pflieger, Sergey I. Nikitenko, Carlos Cairós, Robert Mettin, 2019-01-31 This book presents the latest research on fundamental aspects of acoustic bubbles and in particular on various complementary ways to characterize them It starts with the dynamics of a single bubble under ultrasound and then addresses few bubble systems and the formation and development of bubble structures before briefly reviewing work on isolated bubbles in standing acoustic waves bubble traps and multibubble systems where translation and interaction of bubbles play a major role Further it explores the interaction of bubbles with objects and highlights non spherical bubble dynamics and the respective collapse geometries It also discusses the important link between bubble dynamics and energy focusing in the bubble leading to sonochemistry and sonoluminescence The second chapter focuses on the emission of light by cavitation bubbles at collapse sonoluminescence and on the information that can be gained by sonoluminescence SL spectroscopy e g the conditions reached inside the bubbles or the nature of the excited species formed This chapter also includes a section on the use of SL intensity measurement under pulsed ultrasound as an indirect way to estimate bubble size and size distribution Lastly since one very important feature of cavitation systems is their sonochemical activity the final chapter presents chemical characterizations the care that should be taken in using them and the possible visualization of chemical activity It also explores the links between bubble dynamics SL spectroscopy and sonochemical activity This book provides a fundamental basis for other books in the Molecular Science Ultrasound and Sonochemistry series that are more focused on applied aspects of sonochemistry A basic knowledge of the characterization of cavitation bubbles is indispensable for the optimization of sonochemical processes and as such the book is useful for specialists researchers engineers PhD students etc working in the wide area of ultrasonic processing

Physical Chemistry Robert G. Mortimer, 2000-04-28 This new edition of Robert G Mortimer's Physical Chemistry has been thoroughly revised for use in a full year course in modern physical chemistry In this edition Mortimer has included recent developments in the theories of chemical reaction kinetics and molecular quantum mechanics as well as in the experimental study of extremely rapid chemical reactions While Mortimer has made substantial improvements in the selection and updating of topics he has retained the clarity of presentation the integration of description and theory and the

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Table of Contents Prentice Hall Chemistry Chapter 14 Answers

1. Understanding the eBook Prentice Hall Chemistry Chapter 14 Answers
 - The Rise of Digital Reading Prentice Hall Chemistry Chapter 14 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry Chapter 14 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Chemistry Chapter 14 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry Chapter 14 Answers
 - Personalized Recommendations
 - Prentice Hall Chemistry Chapter 14 Answers User Reviews and Ratings
 - Prentice Hall Chemistry Chapter 14 Answers and Bestseller Lists
5. Accessing Prentice Hall Chemistry Chapter 14 Answers Free and Paid eBooks
 - Prentice Hall Chemistry Chapter 14 Answers Public Domain eBooks

- Prentice Hall Chemistry Chapter 14 Answers eBook Subscription Services
- Prentice Hall Chemistry Chapter 14 Answers Budget-Friendly Options
- 6. Navigating Prentice Hall Chemistry Chapter 14 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Prentice Hall Chemistry Chapter 14 Answers Compatibility with Devices
 - Prentice Hall Chemistry Chapter 14 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Prentice Hall Chemistry Chapter 14 Answers
 - Highlighting and Note-Taking Prentice Hall Chemistry Chapter 14 Answers
 - Interactive Elements Prentice Hall Chemistry Chapter 14 Answers
- 8. Staying Engaged with Prentice Hall Chemistry Chapter 14 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Chemistry Chapter 14 Answers
- 9. Balancing eBooks and Physical Books Prentice Hall Chemistry Chapter 14 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Chemistry Chapter 14 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Prentice Hall Chemistry Chapter 14 Answers
 - Setting Reading Goals Prentice Hall Chemistry Chapter 14 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Prentice Hall Chemistry Chapter 14 Answers
 - Fact-Checking eBook Content of Prentice Hall Chemistry Chapter 14 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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