

NHA STUDY GUIDE

TERM	DEFINITION
THE PROPER WAY TO DISPOSE OF A NEEDLE IS	put it into the sharps container with recapping it, immediately after withdrawing it
THE "GOOD SAMARITAN LAW" ENCOURAGES HEALTHCARE PROFESSIONALS TO	provide medical care within the scope of their training at the scene of an accident without fear of being sued for negligence
IF A PATIENT REFUSES A VENIPUNCTURE PROCEDURE, THE PHLEBOTOMIST SHOULD	immediately report the refusal and actions taken to the nurse or doctor
MICROORGANISMS THAT CAUSE DISEASE	pathogens
THE VEIN IN THE ARM IS MOST SUSCEPTIBLE TO VENIPUNCTURE	median cubital vein
FOOT PRAXIAL	after a meal
TOURQUET'S MAY BE LEFT ON THE PATIENT FOR	2 min
THIS COMPLICATION RESULTS FROM REPEAT ED VENIPUNCTURE OF THE SAME	phlebitis
THE COMPONENTS THAT MAKE UP THE CHAIN OF INFECTION ARE	source, susceptible host, mode of transmission
A HEMATOMA CAN BE PREVENTED IF	pressure is placed on the venipuncture site until the bleeding stops
ARTERIAL BLOOD SAMPLES NEED WHAT KIND OF SPECIAL HANDLING	specimen must be chilled after collection
THE RECOMMENDED DEPTH FOR AN ULNAR MICROCAPILLARY COLLECTION SHOULD NOT EXCEED	2.0 mm
COMMON SYMPTOMS OF SHOCK ARE	pale, cold, clammy skin, rapid weak pulse, unresponsive/lavating eyes
BACTERIA, VIRUSES, FUNGUS OR PARASITES BELONG TO WHICH TYPE OF HAZARD	biologic
EXAMPLES OF PPE	gloves, mask, face shield/goggles
ISOLATION IS CATEGORIZED INTO WHICH THREE CAT EGORIES	airborne, droplet, contact
WHAT IS CONSIDERED A PREANALYTICAL ERROR	allergic reaction
DELIVERS OXYGEN, NUTRIENTS, HORMONES, AND ENZYMES TO THE CELLS AND TO TRANSPORT CELLULAR WASTE SUCH AS CARBON DIOXIDE IS GREAT TO THE ORGANS WHERE THEY CAN BE ELLENED FROM THE BODY	circulatory circulation

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G Psacharopoulos



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 magnesium nitrate and water a balanced equation for this reaction is $\text{MgO} + 2\text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{H}_2\text{O}$ word equation for magnesium
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 type of chemical reaction for this reaction we have a neutralization reaction balancing strategies here we have a
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 solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the
 gaseous hydrogen produced

$\text{HNO}_3 + \text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{H}_2\text{C}_2\text{H}_3\text{O}_2 + \text{Mg}(\text{NO}_3)_2$ chemical equation - Aug 11 2023

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 reaction where one mole of solid magnesium mg and two moles of aqueous nitric acid HNO_3 react to form one mole of

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